

Thursday Sep 18th

ANNOUNCEMENTS

- **TODAY:** Find your **GROUP #** & **SIT** in your **GROUP'S GENERAL AREA** when you come in
- Grades are completed and will be posted in D2L for **Assignment I-1** and **Test #1** later this afternoon.
- Your next RQ (**RQ-3**) will be due **next Tuesday Sept 23**, 30 minutes before class.
- The **Self Test for RQ-3** is now posted. RQ-3 will be available to take immediately after class today.

TOPIC # 5
The RADIATION LAWS
PART 2

Class Notes p 29

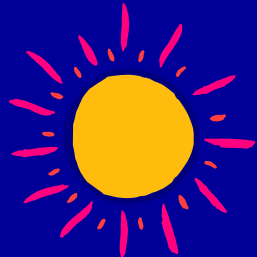
OBJECTIVES:

To understand more
essentials about

Solar radiation

&

Terrestrial radiation



based on the principles of
the last 2 “Radiation Laws.”



THE RADIATION LAWS

**Review of
Laws # 2 – 4**

REVIEW: Match each equation with the correct phrase below & fill in the name of the LAW:

(a) $E = \sigma T^4$ (b) $E = h c / \lambda$ (c) $\lambda_m = a / T$

“The hotter the body, the shorter the wavelength”
The cooler the body, the longer the wavelength”

“The hotter the body, the (much) greater the amount of energy flux or radiation”

“SHORTER wavelengths have HIGHER intensity radiation than LONGER wavelengths”

ANSWERS!

(c) $\lambda_m = a / T$

Wien's Law

“The hotter the body, the shorter the wavelength”
The cooler the body, the longer the wavelength”

(a) $E = \sigma T^4$

Stefan-Boltzmann Law

“The hotter the body, the (much) greater the amount of energy flux or radiation”

(b) $E = h c / \lambda$

Planck Function

“SHORTER wavelengths have HIGHER intensity radiation than LONGER wavelengths”

On to the last two laws

#5 and #6

LAW #5: Radiation & distance

-- the inverse-square law

The inverse square law describes:

how solar **FLUX of ENERGY**
decreases
with **increasing DISTANCE**
from the source of
the radiation flux
i.e., the **Sun**

INVERSE SQUARE LAW =

The amount of radiation passing through a particular unit area is:

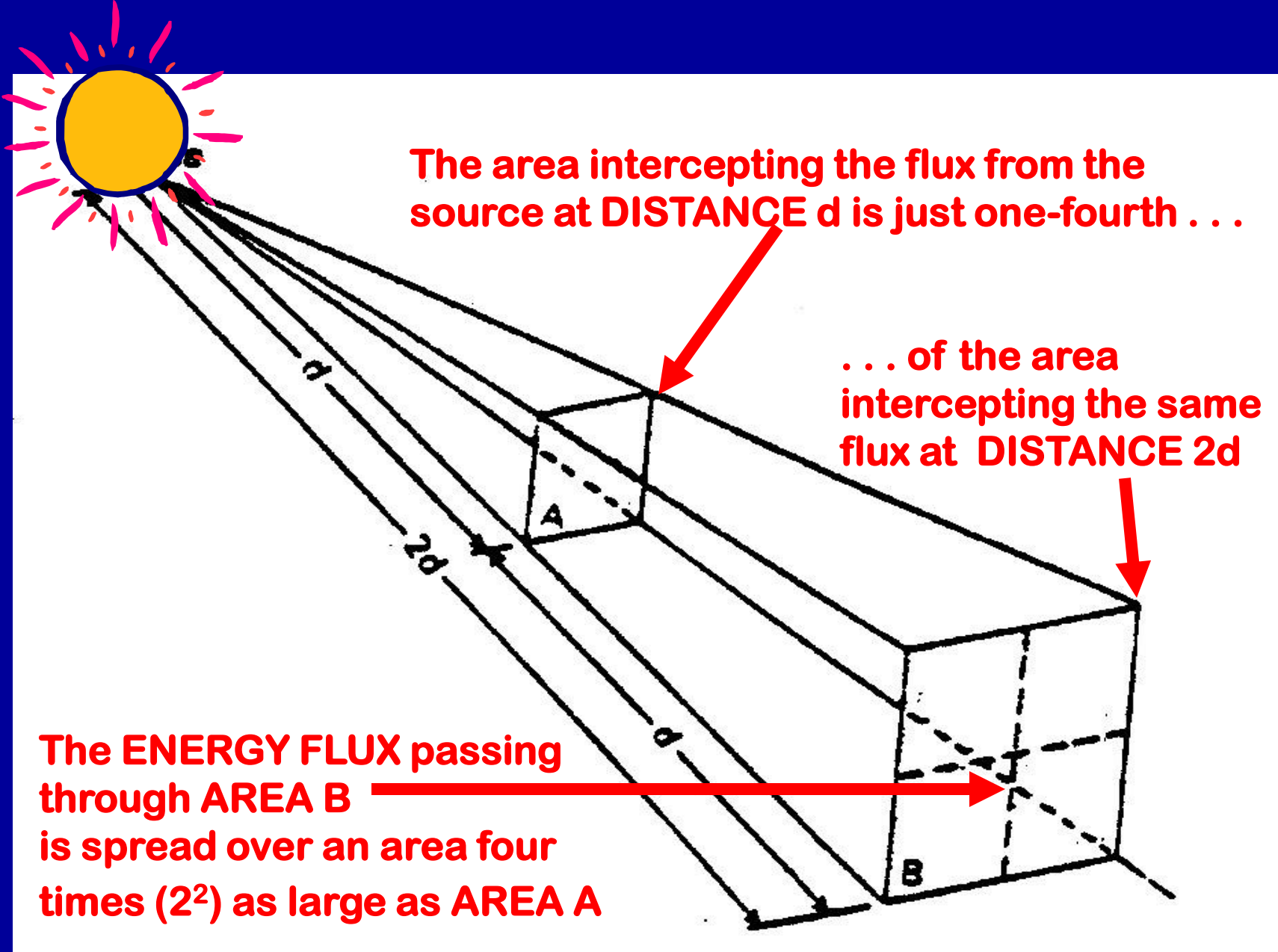
INVERSELY PROPORTIONAL

to the

SQUARE of the distance

of that unit area from the source

$$(1/d^2)$$



Inverse-Square Law (easy way):

If we double the distance from the source to the interception point, the intensity of the radiation decreases by a factor of

$$(1/2)^2 = 1/4$$

OR

If we triple the distance from the source to the interception point, the intensity decreases by a factor of

$$(1/3)^2 = 1/9 \quad \dots \text{etc, etc.}$$



OR

if we reduce the distance from the source to the interception point by a factor of 2 or 3, the intensity of the radiation increases by a factor of

$$2^2 = 4$$

or

$$3^2 = 9$$

... etc, etc.



Why is this concept important?

Because it means that relatively
SMALL changes in distance from
the source of energy
(e.g., the Sun)

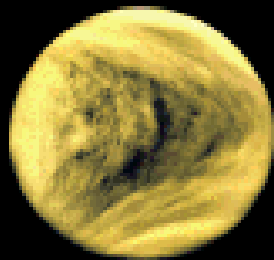
can result in LARGE changes in the
amount of energy received
by a planet's surface.



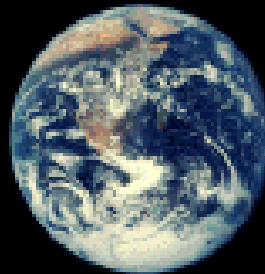
GOLDILOCKS & THE 3 PLANETS



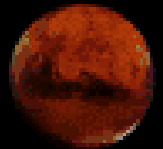
 To
Sun



VENUS



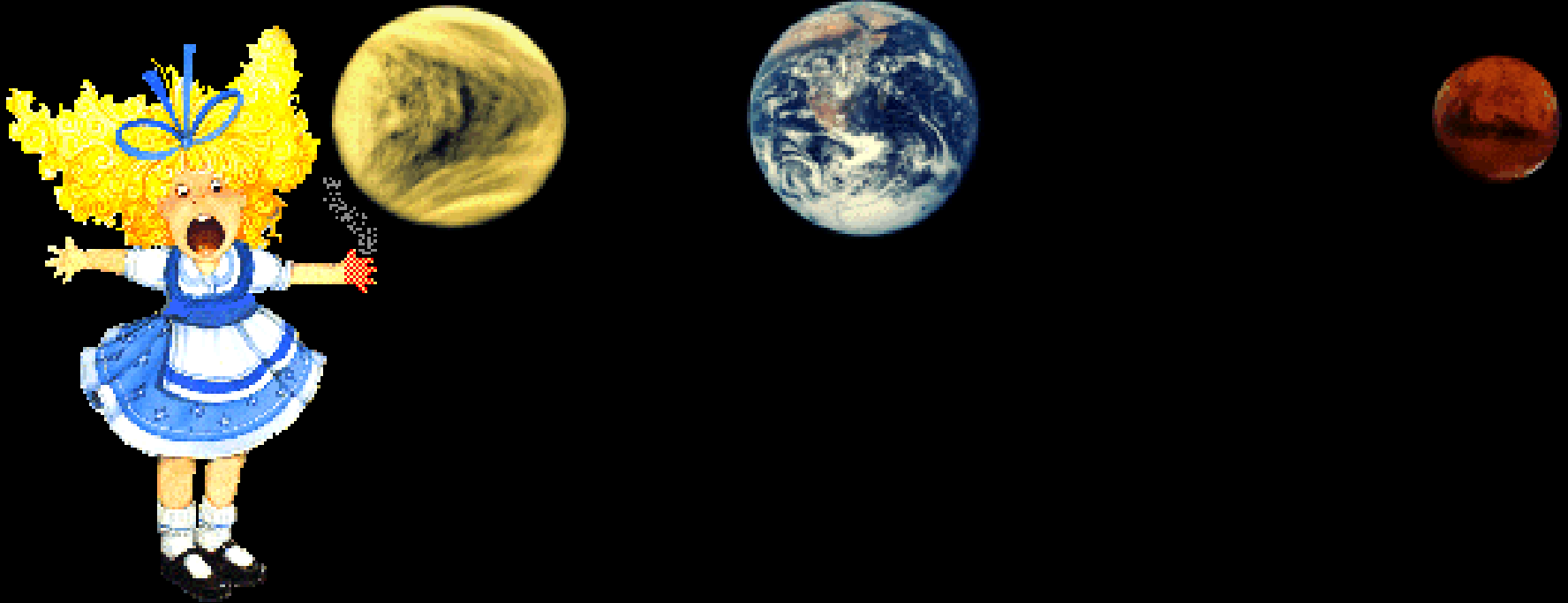
EARTH



MARS



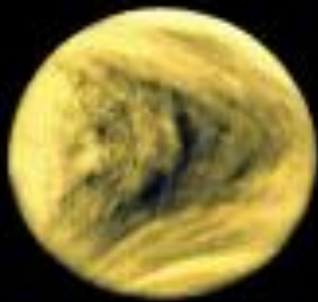
GOLDILOCKS & THE 3 PLANETS



Yikes! Venus is too HOT!



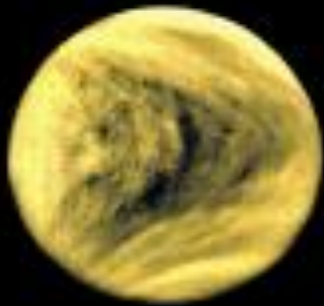
GOLDILOCKS & THE 3 PLANETS



Brrrrrrrrr, Mars is too COLD!!



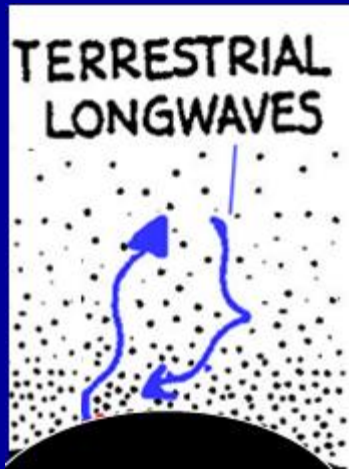
GOLDBLOCKS & THE 3 PLANETS



Ahhhh! Earth
is **JUST RIGHT!**

But is being at “just the right distance”
the primary determinant of
Earth’s temperature?





The **absorption** and re-radiation of Infrared radiation by **GH Gases** . . .

. . . is what keeps the Earth in the **"just right" temperature range** for water to be present in all 3 phases . . .
. . . and just right for US too!



Without the "Greenhouse Effect"
the Earth would be TOO COLD
for life as we know it!



**Thanks,
Greenhouse
Effect!**



Q1 The inverse-square law (when applied to the distance between a planet and the Sun) is all that is needed to determine that planet's temperature.

YES or NO?

1. Yes, this is what the **Goldilock's Effect** is illustrating.
2. No, how much solar energy the planet **reflects back** must also be taken into account
3. No, whether or not the planet has a **greenhouse effect** must also be taken into account.

Q1 The inverse-square law (when applied to the distance between a planet and the Sun) is all that is needed to determine that planet's temperature.

YES or NO?

1. Yes, this is what the **Goldilock's Effect** is illustrating.
2. No, how much solar energy the planet **reflects back** must also be taken into account
3. No, whether or not the planet has a **greenhouse effect** must also be taken into account.

Both 2 & 3
are correct!

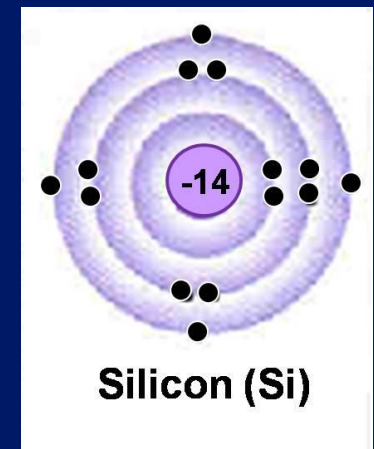
**Re-Read SGC
p 43 (look for
the 2nd green
Pushpin note!)**



Yay! Another Sustainability Segment!



Starring:



<http://www.pbs.org/wgbh/nova/tech/saved-by-the-sun.html>

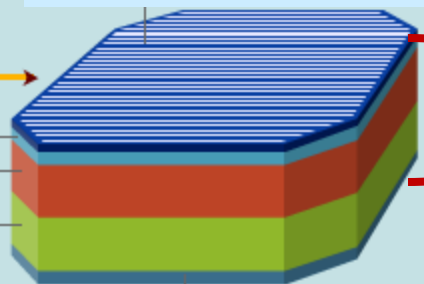
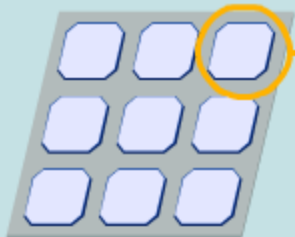
INSIDE A SOLAR CELL

SOLAR PANEL

PHOTOVOLTAIC CELL (PV)

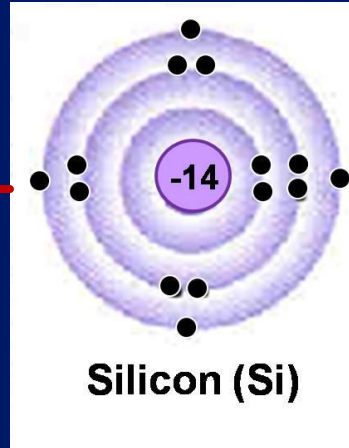
Antireflective coating

metal conducting strips

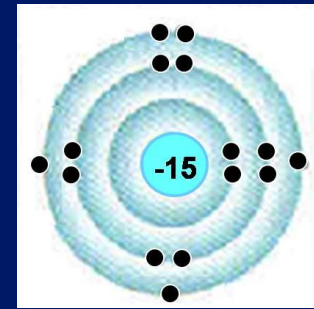


metal backing

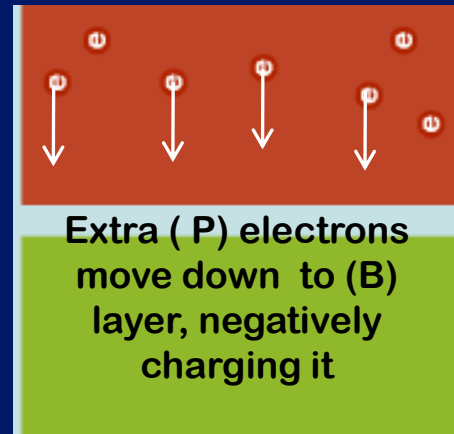
Silicon Layers



Silicon (Si)

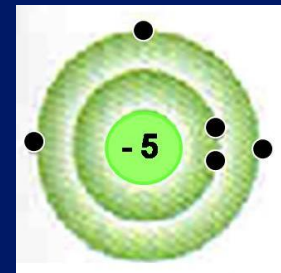


Phosphorus (P)
“doped” Si layer



Extra (P) electrons
move down to (B)
layer, negatively
charging it

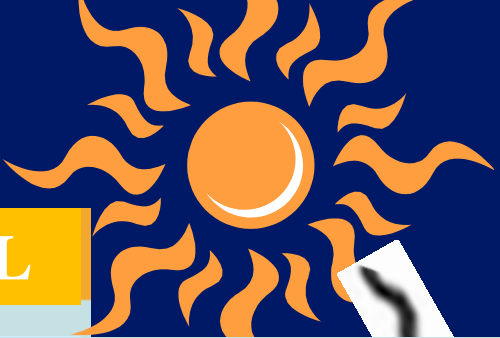
Boron (B)
“doped” Si layer



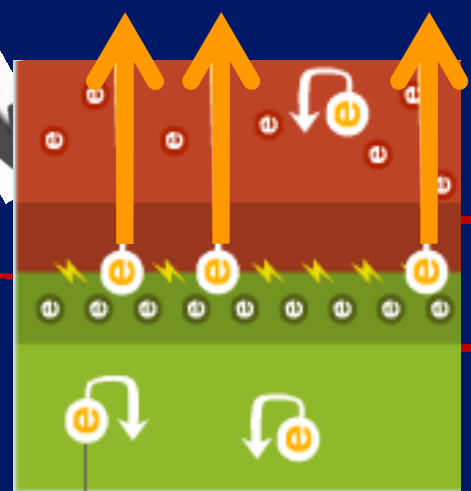
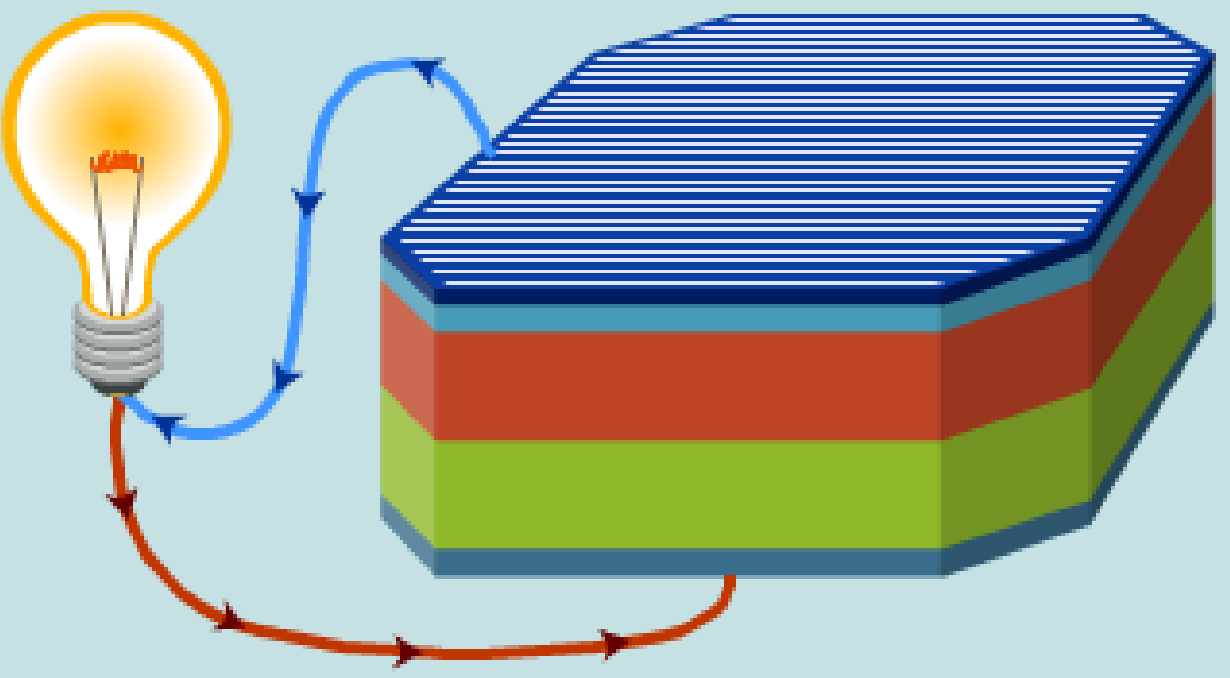
Read this
explanation at:

[http://www.pbs.org/wgbh/nova/tech/how-solar-cell-works.html /](http://www.pbs.org/wgbh/nova/tech/how-solar-cell-works.html/)





INSIDE A SOLAR CELL



ELECTRIC FIELD



BEFORE



The Maddox-Hirschboeck Residence

AFTER



16 SunPower
230 watt
modules

3.68 kW @ 156 hrs
Estimated Monthly
Production: 576 kWh

6,912 kWh
annually





Net Meter & Electric Panel



Inverter
(DC → AC)
Solar Meter
& AC
Disconnect



SUNPOWER MONITORING

Lifetime:
25,1377 kWh
SOLAR GENERATED

ENVIRONMENTAL SAVINGS:



40,170 lbs

Total CO₂ emissions avoided



Equivalent to

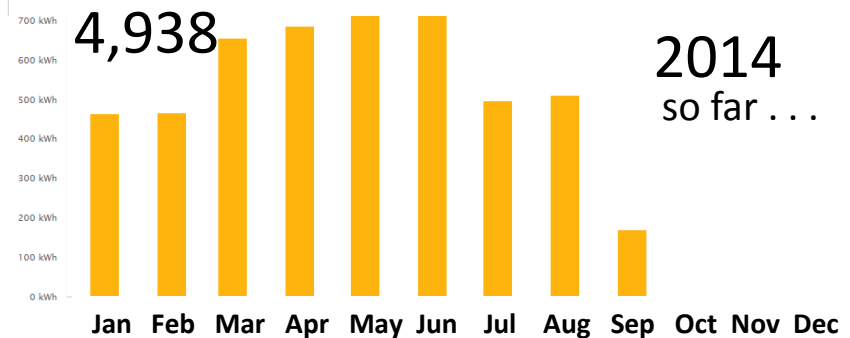
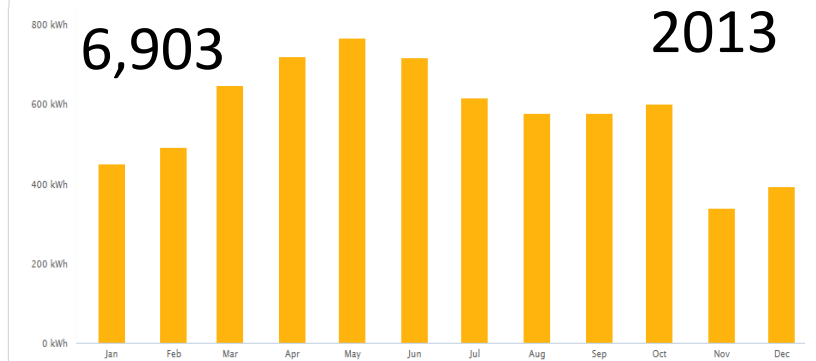
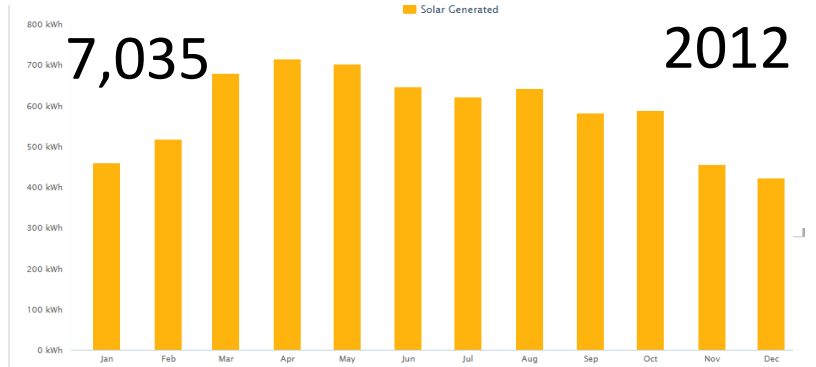
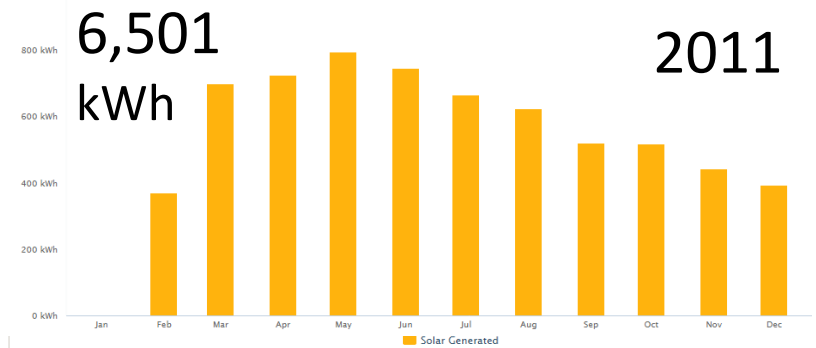
41,961 miles not driven

&



467 trees grown

for 10 years



Going Electric: My LEAF "EV"



Electric
&
Solar
powered!



"plugging in"
at home

NATIONAL DRIVE ELECTRIC WEEK - TUCSON / TUCSON PLUGS IN 2014

SUNDAY, SEPTEMBER 21, 2014, 8:00AM - 1:00PM

Location: Bookman's Sports Exchange
3330 E. Speedway Blvd.
Tucson, AZ 85716



Expected Plug-In Vehicles

Vehicle	Registered
Nissan LEAF	5
Tesla Model S	3
Ford C-Max Energi	1
Other Plug-In Vehicle	1
Tesla Roadster	1
Zero Motorcycle	1
6 Models	12

Registered attendees report 212,041 electric miles driven.

Also this SUNDAY

SEPTEMBER 21, 2014

A massive, history-making march in New York City.
Hundreds of coordinated actions around the world.



[Tucson Peoples' Climate March - Facebook](#)



PERFECT TIMING

World leaders will be gathered in NYC for a landmark U.N. climate meeting – just the right moment for big public pressure.



MASSIVE SCALE

We'll peacefully flood the streets in historic numbers, both in New York City and in solidarity events around the world.



UNPRECEDENTED COLLABORATION

Over **1,400** (!) businesses, unions, faith groups, schools, social justice groups, environmental groups and more, all working together.



CENTERED ON JUSTICE

Committed to principles of environmental justice and equality – representing the communities that are being hit the hardest by climate change.

and beginning Tuesday . . .



Welcome to the United Nations. It's your world.

عربي 中文 English Français Русский Español

CLIMATE SUMMIT 2014

CATALYZING ACTION

UN HEADQUARTERS · NEW YORK
23 SEPTEMBER 2014

Home About FAQs News Programme Action Areas Thematic Sessions Media Participate Contact

THE LAST LAW!

Law #6:

**“Selective Absorption
and Emission”**

... an important aspect of Law #6 is:

Electromagnetic energy does not NEED matter to be transferred, but when it DOES **react with matter**, it can be:

○ **ABSORBED (and EMITTED)**

○ TRANSMITTED

○ SCATTERED, or

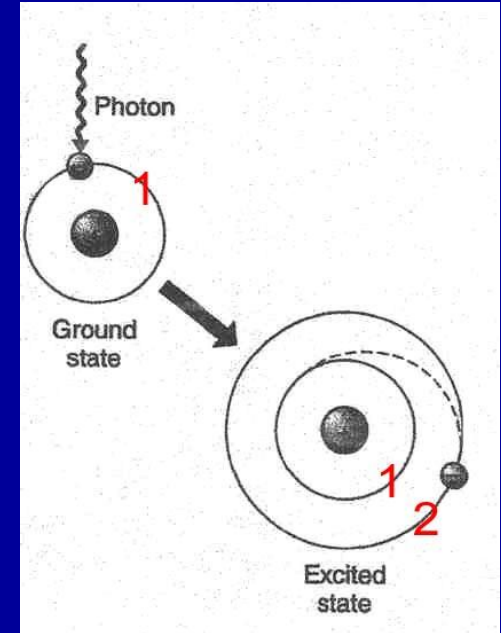
○ REFLECTED ...

... through -- or by -- the matter

So lets review some slides
from previous classes related to
electromagnetic energy,
absorption & emission ...

Do you remember this from Topic # 4 ?

ELECTRONS can make “quantum” leaps between the orbits (or energy shells) by **ABSORBING** or **EMITTING** exactly the **energy difference** between the orbits



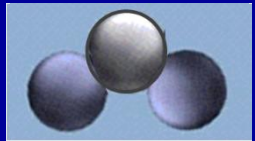
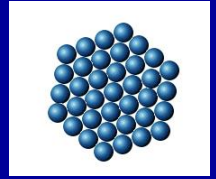
But how? . . . It happens when the electron **absorbs** or **emits** exactly the right **wavelength** & **frequency** of electromagnetic energy for that particular atom

Review – just listen 😊

Quantum behavior also takes place in molecules :

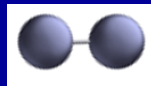
GAS MOLECULES

are able to **absorb** or **emit** only those
frequencies & wavelengths
of electromagnetic energy that
“match up” with a molecule’s
frequencies of vibrating,
bending,
or rotating

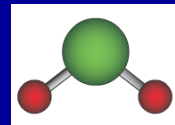


. . . . These **frequencies** depend on
different aspects of the molecule’s
structure & bonding between its atoms:

*For
example:*



“di-atomic”



“tri-atomic”

Just
listen 😊



So . . . the LINK to GLOBAL CHANGE is . . .

Infrared photons absorbed by CO₂

Infrared photons emitted by CO₂

The COMET Program

Infrared photon is emitted →

Faster rotation rate **H₂O** Slow rotation rate

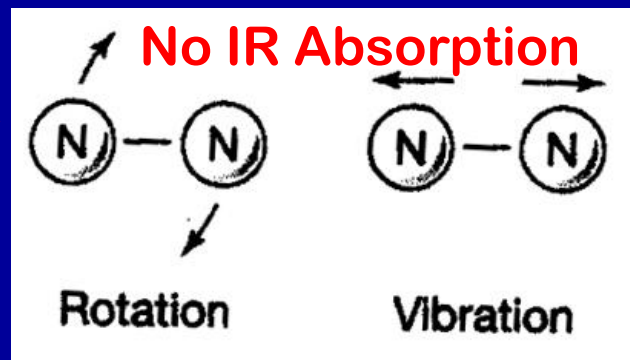
Tri-atomic!

The type and frequency of molecular quantum motions in gases like **CARBON DIOXIDE** and **WATER VAPOR** explain why THEY contribute to **The Greenhouse Effect** while other gases (**O₂, N₂ . . .**) do not!!

Di-atomic!

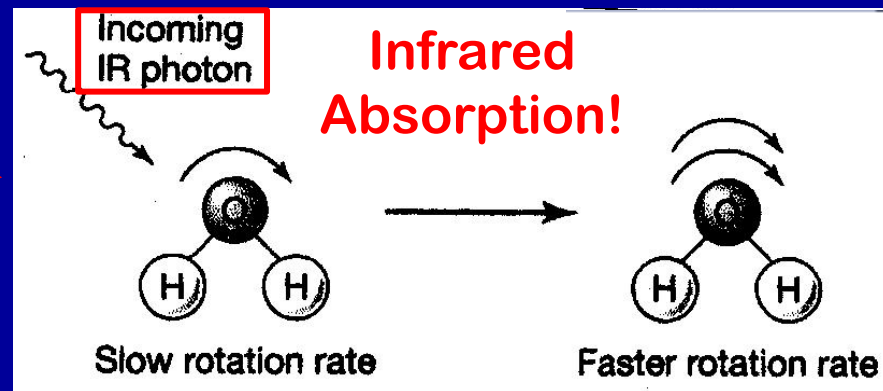
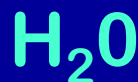
N_2 is NOT a Greenhouse gas:

NITROGEN GAS
MOLECULE

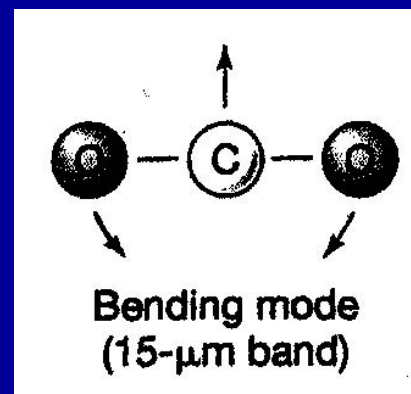


BUT H_2O and CO_2 ARE Greenhouse gases:

WATER VAPOR
MOLECULE

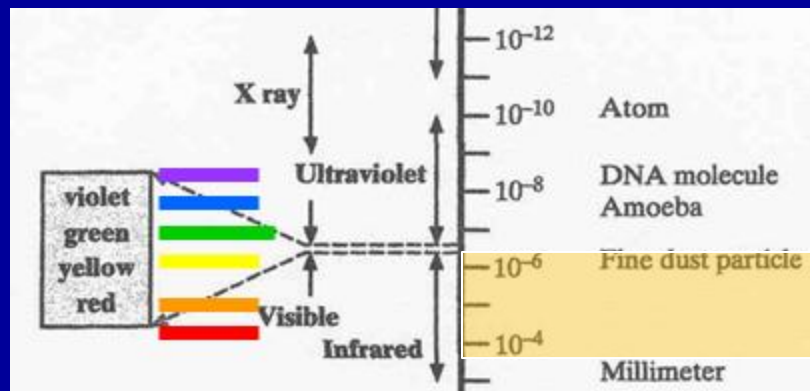


CARBON
DIOXIDE GAS
MOLECULE



Review of the definition: Greenhouse Gas

GHG = a gas than can **absorb** and **emit**
INFRARED wavelengths
of Electromagnetic Radiation



0.7 to 1000+ micrometers

IR radiation

LAW #6: Selective emission and absorption (of gases)

has 2 parts:

a) Some substances emit and absorb radiation at **certain wavelengths only.**

(This is mainly true of gases.)

b) These substances **absorb only** radiation of **wavelengths they can emit.**

Two implications of Law #6:

“ IR absorbed by the gas → IR emitted by the gas ”

[The **frequency & wavelength** of energy absorbed by a particular gas molecule **will be the same as** the frequency & wavelength with which it is emitted.]

&

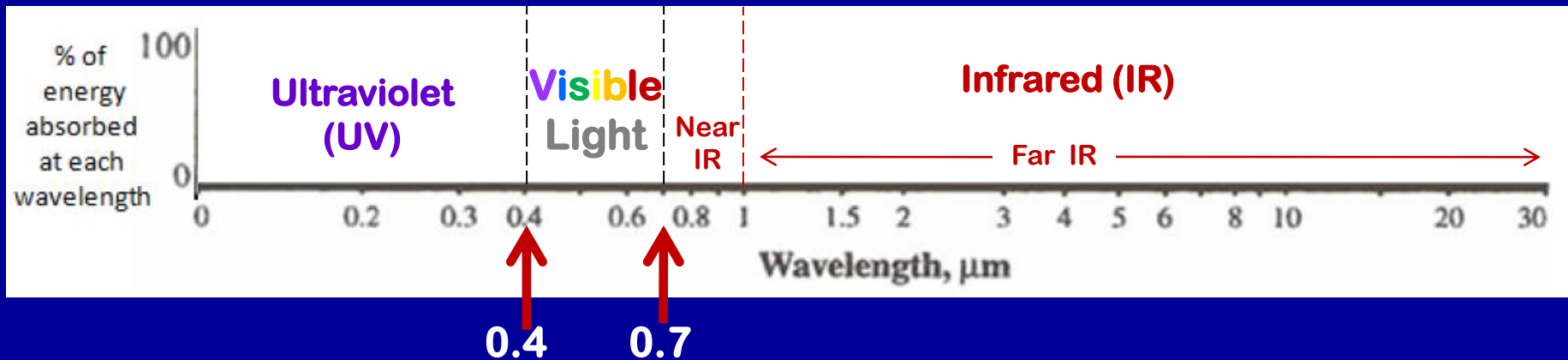
“ IR NOT absorbed = IR transmitted ”

[Wavelengths of energy that are NOT absorbed (or only partially absorbed) by a gas molecule, **get transmitted** right through the ATMOSPHERE!]

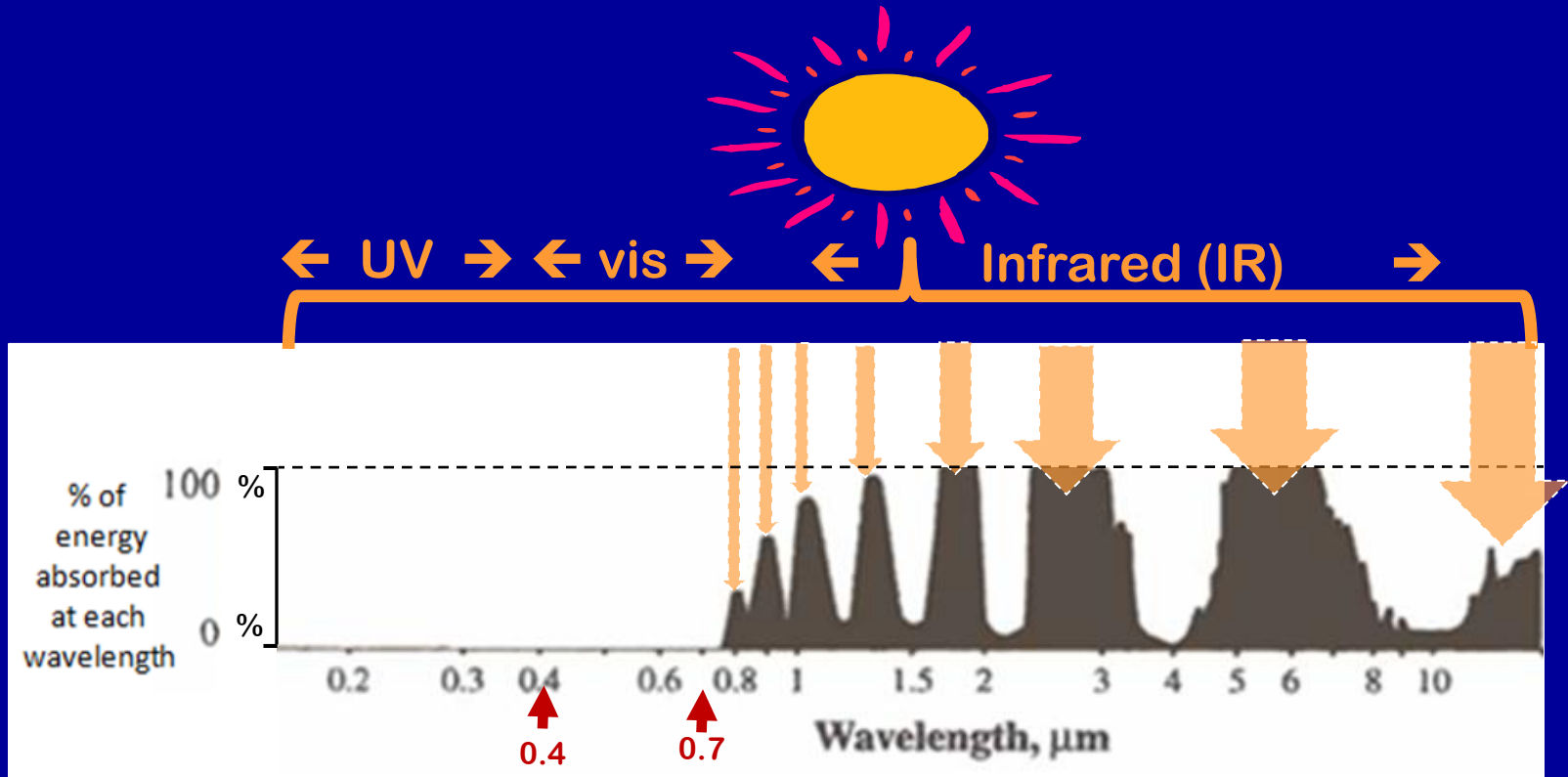


So let's tie Law #6 to the Electromagnetic Spectrum

Review of the spectrum:



The pattern of electromagnetic wavelengths that are **absorbed (& emitted)** by a particular gas molecule . . . is called the gas's **Absorption Spectrum** or **ABSORPTION CURVE**



Radiation is **ABSORBED** (or partially **ABSORBED**) at THESE wavelengths by this particular gas!



But is **ABSORPTION & EMISSION** ALL that happens to Electromagnetic Energy?

Electromagnetic energy does not NEED matter to be transferred, but when it DOES **react with matter**, it can be:

- **ABSORBED (and EMITTED)**

- **TRANSMITTED**

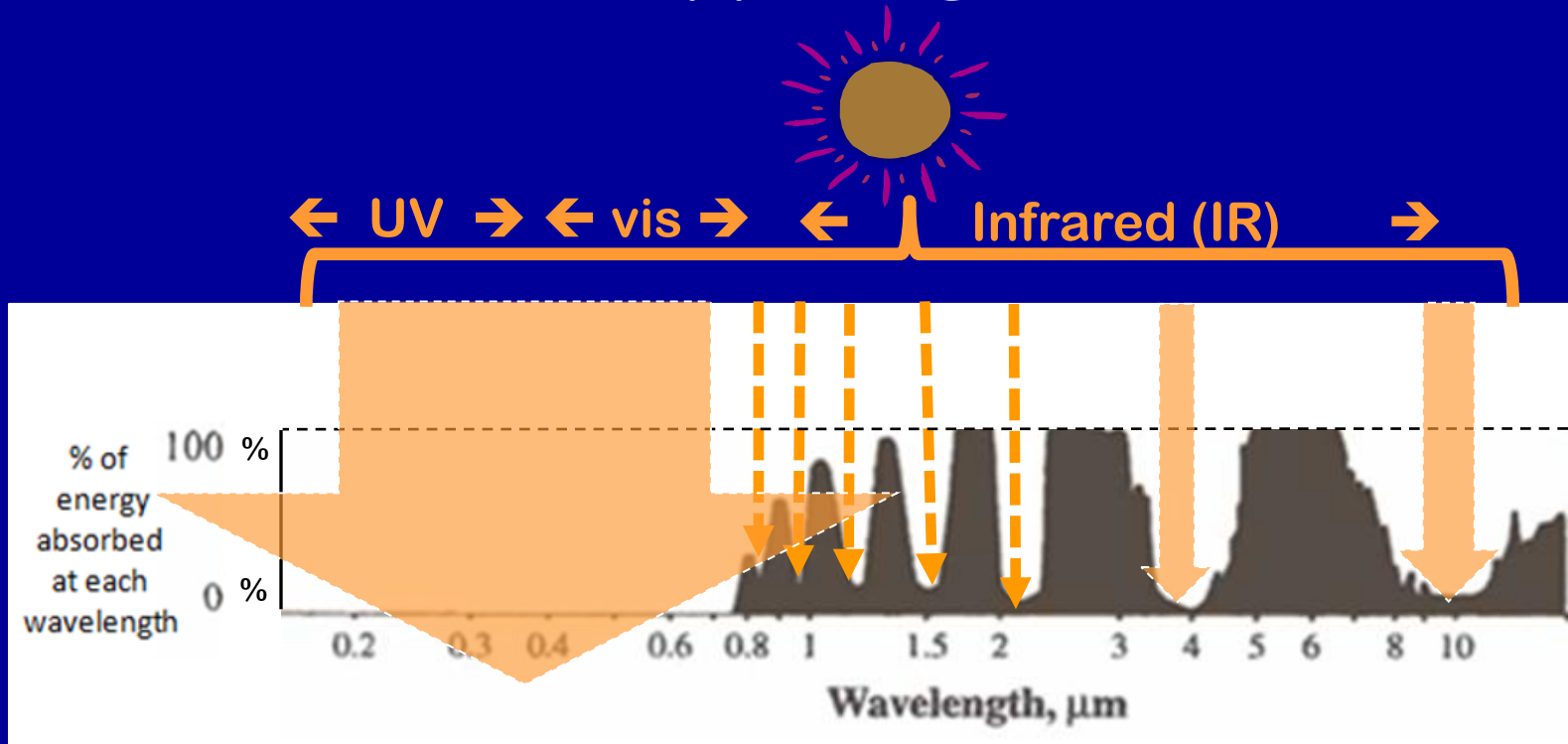
- SCATTERED, or

- REFLECTED . . .

. . . . through -- or by -- the matter

Here's the absorption curve of the same gas:

Radiation is **TRANSMITTED** through the atmosphere at the wavelengths where **no absorption** is happening!

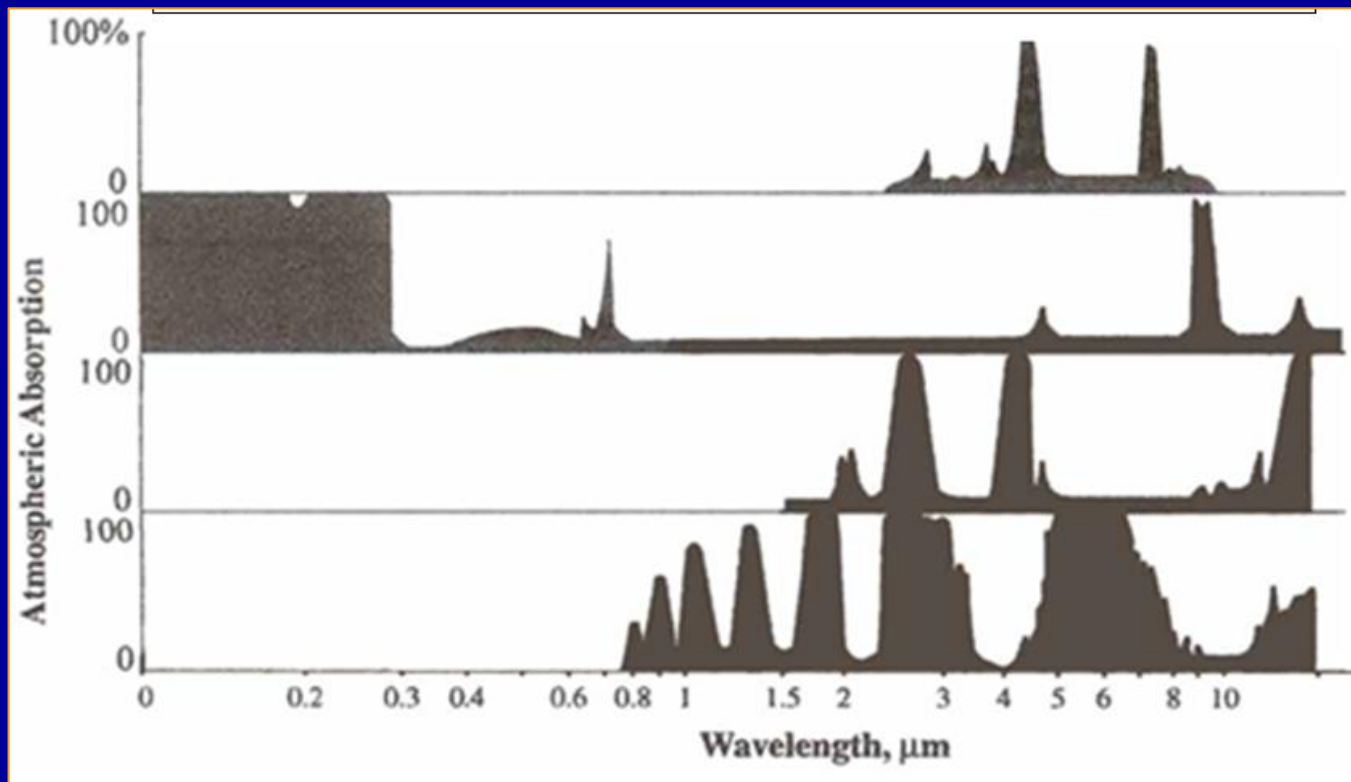


[The “open areas” (0 % or low values) on the **Absorption Curve** represent electromagnetic wavelengths that are **NOT absorbed** (or only partially absorbed) by a particular gas molecule .]



Here are **Absorption Curves** for 4 different gases:

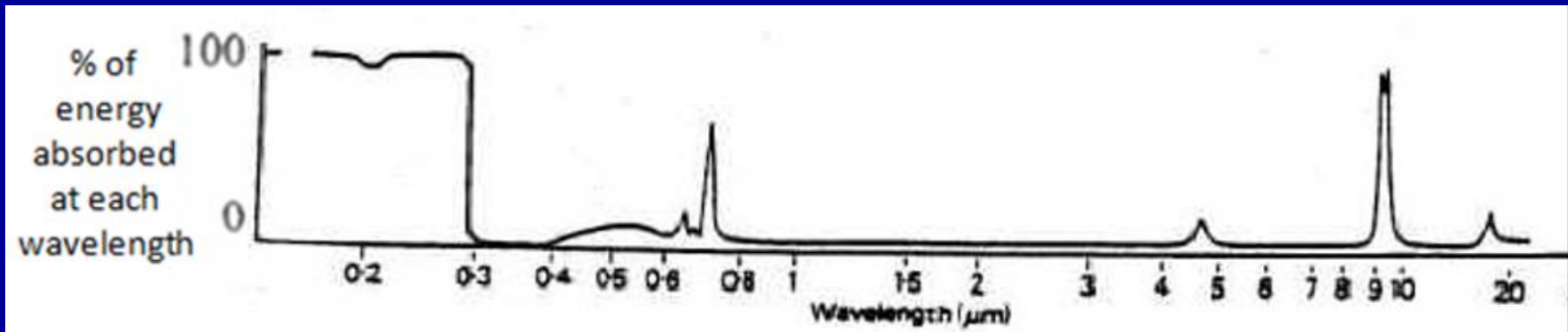
energy
absorbed
at each
wavelength



Each row is
the **Curve**
for a
different gas



Here's another view of an
absorption curve
(without shading under the curve)



ABSORPTION CURVES

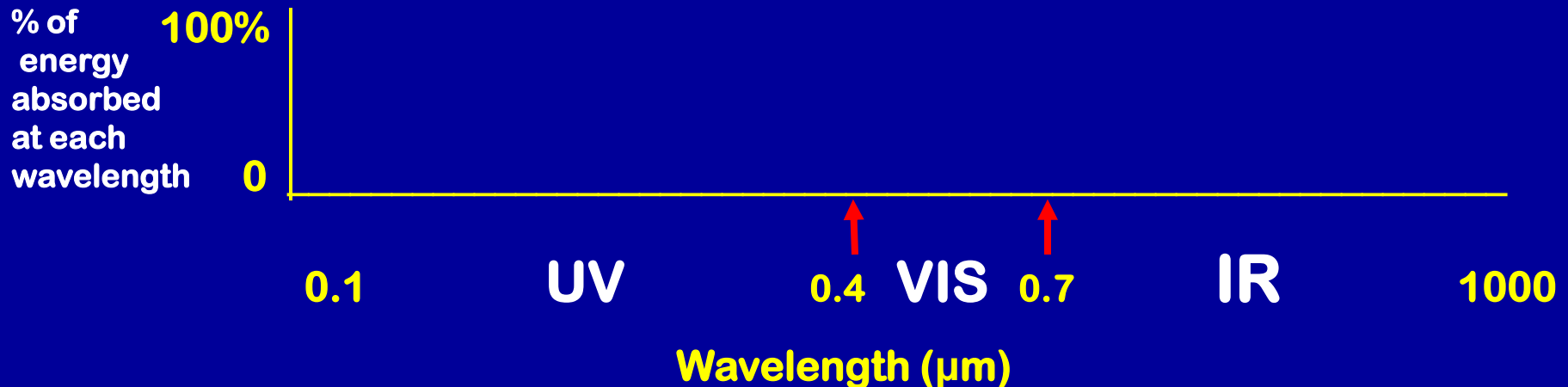
SKETCH THE AXES IN

(on the bottom of p 29 or in your own notes)



HORIZONTAL AXIS: **wavelength**

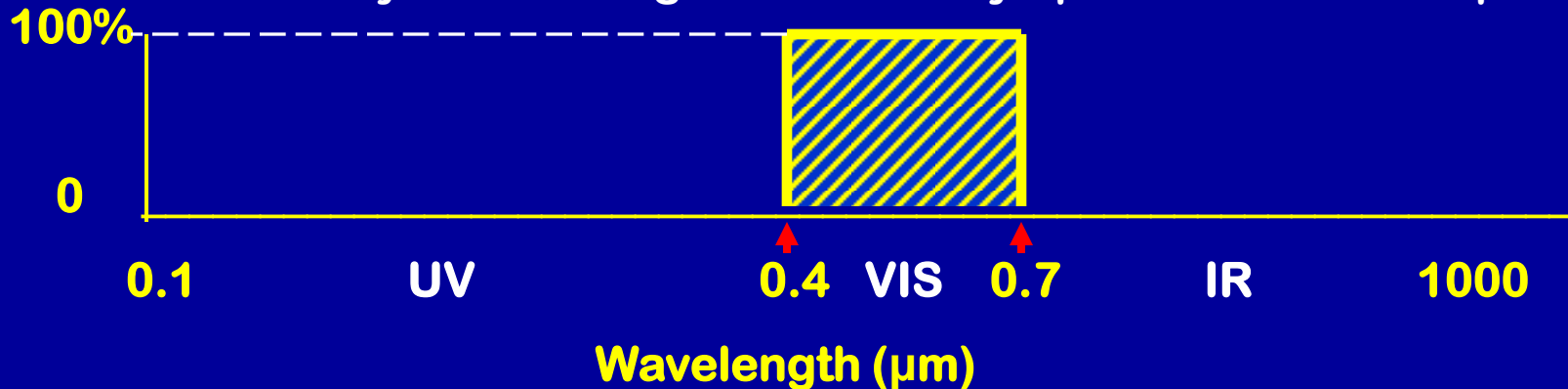
VERTICAL AXIS: **% of energy** at each wavelength
that is **ABSORBED**



What would a curve for a hypothetical gas that absorbs **ALL VISIBLE LIGHT** but **ZERO UV** or **IR**

LOOK LIKE ??

Be sure your sketch goes all the way up to 100% for this question!



SKETCH IT IN



And now

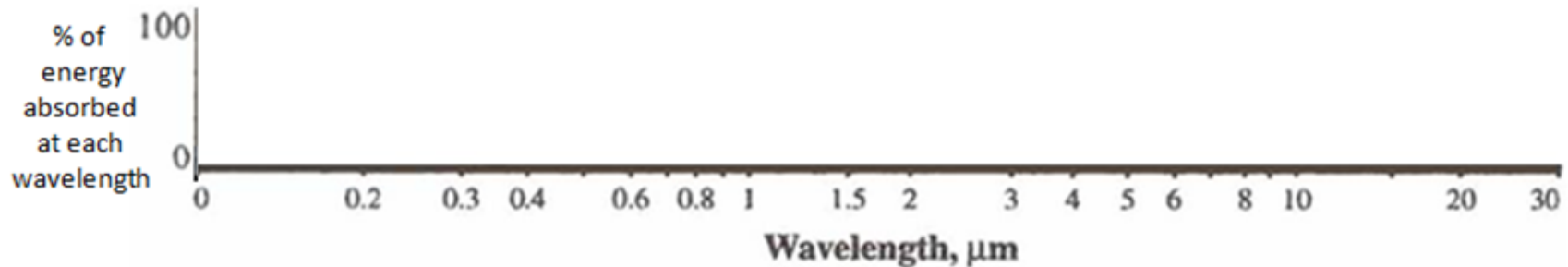
GROUP ASSIGNMENT G-1

**Understanding Radiation,
Absorption & Wavelengths
of the
Electromagnetic Spectrum**

WORTH 10 pts

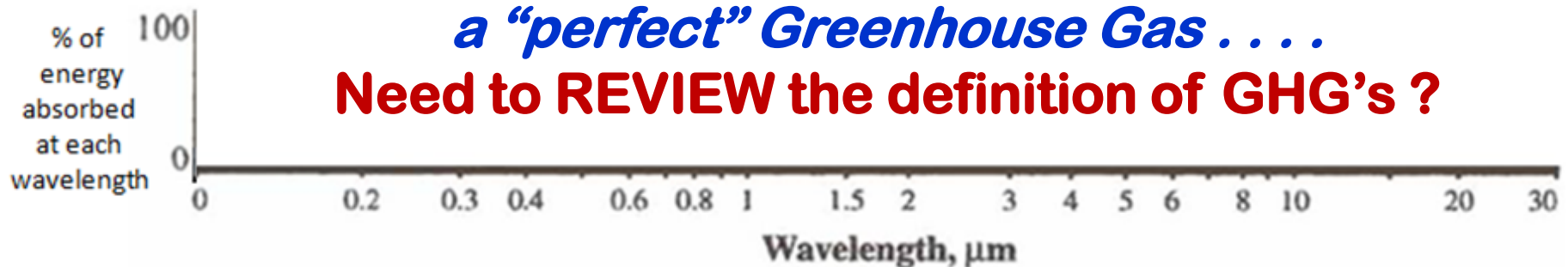
Q1. All UV but zero vis and zero IR?

Q1. Draw an absorption curve for a hypothetical gas that can absorb ALL UV radiation but zero visible light and IR radiation. Then shade in the area under your curve in this and subsequent questions.



Q2. All IR but zero vis and zero UV?

Q2. Draw an absorption curve for a “perfect” greenhouse gas that absorbs ALL IR radiation, but no visible or UV:

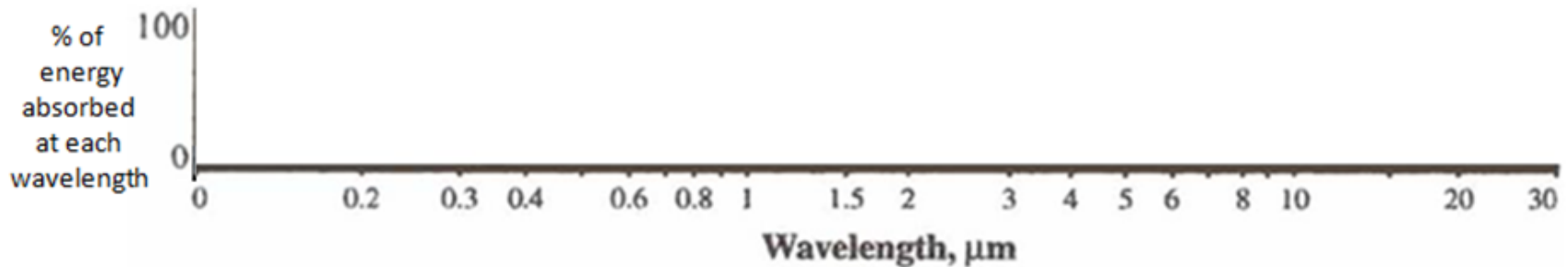


*Remember the GHG definition
to help you answer Q2:*

**Greenhouse gases are gases which
both absorb and emit
electromagnetic radiation in the
infrared (IR) part of the spectrum.**

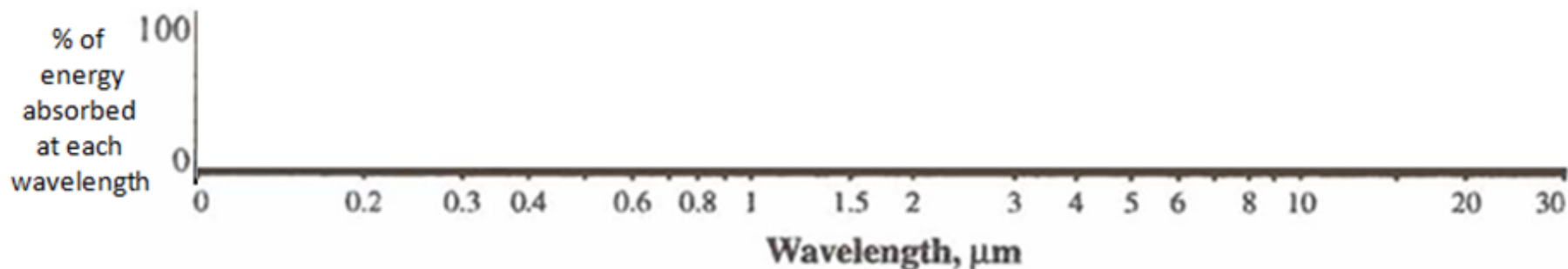
Q3. All UV & IR absorbed but VIS transmitted?

Q3. Draw an absorption curve for a hypothetical gas that absorbs ALL UV radiation and ALL IR radiation, but leaves a "WINDOW" open for visible light, allowing the visible light wavelengths to pass through the gas unimpeded without being absorbed:



Q4. All IR absorbed in specific wavelength bands?

Q4. Draw an absorption curve for a hypothetical gas that can absorb 100% of the IR radiation in these three wavelength bands: band from 2 to 2.5 μm band from 3 to 4 μm band from 13 to 20 μm



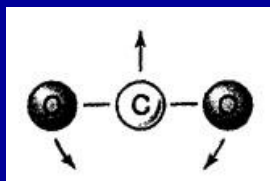
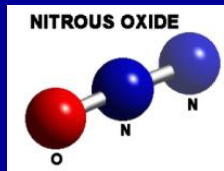
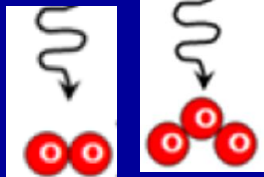
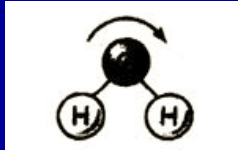
Q5. Is the hypothetical gas in Q4 likely to be a GREENHOUSE GAS?

YES NO *(circle one)*

Briefly explain WHY you answered YES or NO: *(in a few sentences)*

(discuss in your group first!)

Q6 is on p 32



Gas	Here are the specific wavelengths each gas absorbs!	Primary absorption wavelengths (in micrometers)
------------	--	--

Water vapor (H₂O)	0.8	4 to 7
	1	9 to 10
	1.5	11 to 20
	2 to 3.5	

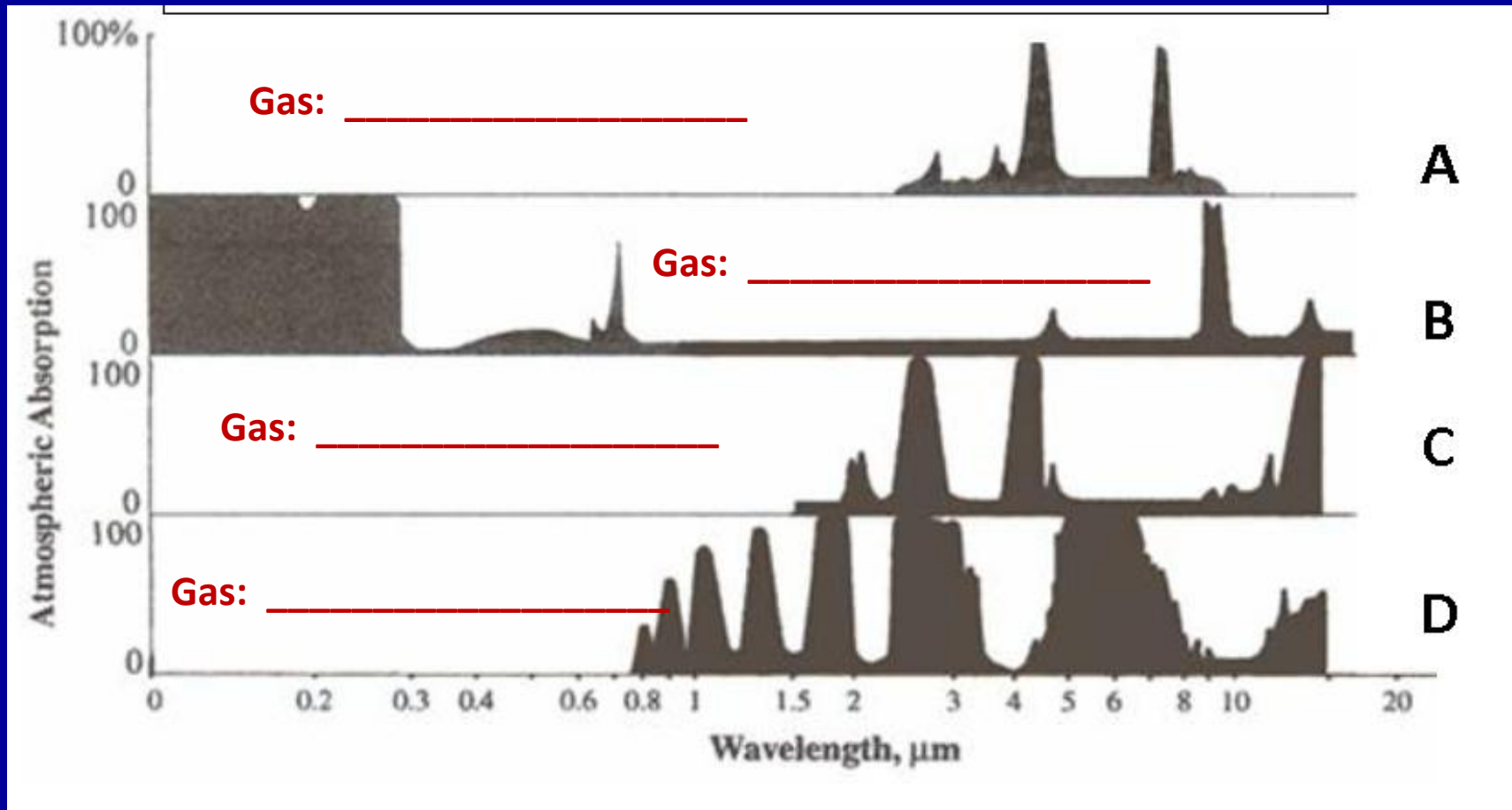
Molecular oxygen (O₂) and Ozone (O₃)	0.0001 to 0.280	
	8.5 to 10	

Nitrous oxide (N₂O)	4 to 5	
	7 to 7.5	

Carbon dioxide (CO₂)	2 to 2.5	
	3 to 4	
	13 to 20	

Match the GAS with its Absorption Curve:

CHOICES: CO_2 H_2O $\text{O}_2 + \text{O}_3$ N_2O



Solar vs. Terrestrial Radiation

Class Concepts Self Test

For Q7 – Q11
work individually or in
pairs on the last page
(same as p 33 in CLASS NOTES)

.... then compare
answers with the rest of
the Group and record
the group's consensus
answers on the G-1 form.

SOLAR vs TERRESTRIAL RADIATION CLASS CONCEPTS SELF TEST

KEY:
= represents Solar shortwave (SW) radiation
= represents Terrestrial longwave (LW) (infrared IR radiation)
= represents the atmosphere and its gases (which can absorb and emit certain kinds of radiation)

Q1. Which diagram above shows SW radiation being reflected back to space?
Diagram A Diagram B Diagram C None of them

Q2. Diagram A shows LW radiation "bouncing off" the gases in the atmosphere (i.e. being reflected back to the surface by the gases without being absorbed by them.) Is this an accurate depiction of how the Greenhouse Effect works? Yes No Partly

Q3. Diagram B shows LW radiation being absorbed and then emitted by the gases in the atmosphere. Is this an accurate depiction of how the Greenhouse Effect works? Yes No Partly

Q4. Diagram C shows LW radiation going right through the atmosphere out to space. Is this an accurate depiction of how the Greenhouse Effect works? Yes No Partly

Q5. On the diagram that you think best depicts the processes involved in the GREENHOUSE EFFECT, circle the specific part of the diagram that represents the Greenhouse Effect.

Q6. Below is a modified version of the cartoon. It is more complete and more accurate, but there are still some important processes not being represented. Can you think of what they might be?

Modified Cartoon of Solar (SW) & Terrestrial (LW / IR) wavelengths of radiation:

① Some incoming SW radiation from the SUN goes right through the atmosphere to Earth (w/o being absorbed)

② The Earth absorbs SW that reaches the surface

③ Some IR radiation is emitted from the Earth's surface right out to space through "IR window"

④ Some IR radiation is absorbed by GH gases in the atmosphere and emitted back to Earth

⑤ Some IR radiation is absorbed by GH gases in the atmosphere, but is emitted out to space (not back to Earth)

Will be a good review of Topic 5!

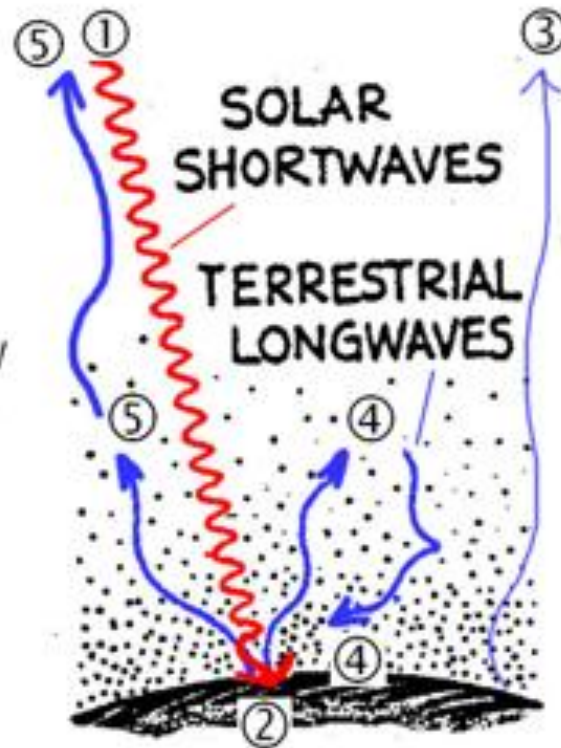
THINKING MORE DEEPLY “TAKE HOME”

Modified Cartoon of Solar (SW) & Terrestrial (LW / IR) wavelengths of radiation:

① Some Incoming SW radiation from the SUN goes right through the atmosphere to Earth (w/o being absorbed)

② The Earth absorbs SW that reaches the surface

③ Some IR radiation is emitted from the Earth's surface right out to space through “IR window”



④ Some IR radiation is absorbed by GH gases in the atmosphere and emitted back to Earth

⑤ Some IR radiation is absorbed by GH gases in the atmosphere, but is emitted out to space (not back to Earth)

This diagram is more complete and more accurate than the one at the top of the page, but there are still some important processes not being represented.

Can you think of what they might be?

When you get
your **GROUP
FOLDER . . .**



. . . You'll find your **GROUP ANSWER FORM:**

★ GROUP ASSIGNMENT G-1: UNDERSTANDING RADIATION, ABSORPTION & WAVELENGTHS OF THE ELECTROMAGNETIC SPECTRUM (worth 50 pts)

GROUP # _____

Each group participant's SIGNATURE: _____

GROUP LEADER: _____

GROUP LEADERS: _____

When your group is done, check to be sure that each member of the group who participated today has signed the top of the form.
Then insert your G-1 form into your group folder and return the folder to the front of the classroom.

BACKGROUND (Radiation Law #6):

ABSORPTION CURVES (diagrams that show which wavelengths of energy different gases selectively absorb)

We use an absorption curve (graph) to show the relationship between wavelengths of the electromagnetic spectrum (along the horizontal axis) and the % of energy at each wavelength that is absorbed by a particular gas (vertical axis).

Q1. Draw an absorption curve for a hypothetical gas that can absorb ALL UV radiation but 20% visible light and in addition, then shade in the area under your curve in this and subsequent questions.

% of energy absorbed at each wavelength

0 0.1 0.2 0.3 0.4 0.5 0.6 0.7 0.8 0.9 1 1.1 1.2 1.3 1.4 1.5 1.6 1.7 1.8 1.9 2 2.1 2.2 2.3 2.4 2.5 2.6 2.7 2.8 2.9 3 3.1 3.2 3.3 3.4 3.5 3.6 3.7 3.8 3.9 4 4.1 4.2 4.3 4.4 4.5 4.6 4.7 4.8 4.9 5 5.1 5.2 5.3 5.4 5.5 5.6 5.7 5.8 5.9 6 6.1 6.2 6.3 6.4 6.5 6.6 6.7 6.8 6.9 7 7.1 7.2 7.3 7.4 7.5 7.6 7.7 7.8 7.9 8 8.1 8.2 8.3 8.4 8.5 8.6 8.7 8.8 8.9 9 9.1 9.2 9.3 9.4 9.5 9.6 9.7 9.8 9.9 10

Group Assignment grades
are given to those
students **present in class**
who work on the activity
and who indicate this by
SIGNING THE FORM . . .

G-1 Group Assignment Form

Enter your **GROUP #** → GROUP #

Each Group Participant's SIGNATURE:

PRINT NAME legibly next to the signature:

GROUP LEADER: Stanley Student

Stan Student

Stella Student

STELLA STUDENT


**Your
SIGNATURE**


**Your name
PRINTED**

**BOTH
your
signature
& printed
name are
needed →**

TODAY'S GROUP LEADER = Student whose last name is LAST in the ALPHABET on the Group List
LEADER passes the GROUP FORM around so each student SIGNS IT & PRINTS HIS /HER NAME

**OK – LETS
GET TO WORK!**

- **REMEMBER:** Please **do not remove** the **GROUP FOLDERS** from the **CLASSROOM**
- **REMINDER:** **RQ-3** is due **Tue Sep 23**
(30 minutes before our next class)

- Want to get active and do something about **Climate Change?**

SUNDAY Sep 21 – Join the rest of the nation in the **People's Climate March**

[Tucson Peoples' Climate March - Facebook](#)

and / or visit **Tucson Plugs In – 2014**

<http://tucsonelectricvehicle.org/events.html>