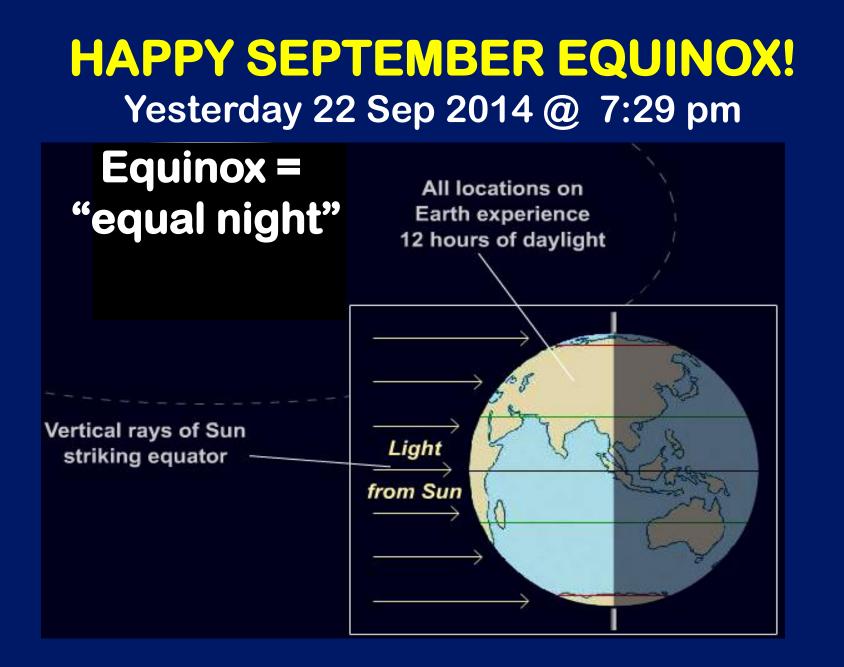
Tuesday Sep 23th SIT IN YOUR GROUP AREA TODAY! Wrap-Up of Topic #5 on the Radiation Laws to get ready for Topic # 6 Atmospheric & Structure & Composition

• RQ-3 was cutoff 30 minutes before class today – if you missed it see FAQ #22

Clickers today – get ready!

Your Clicker Participation Percentage (%) is now visible in D2L in your GRADEBOOK. [If you've been clicking, but have 0%, your points are saved, but they are NOT yet linked to your name . . . so REGISTER and you will see them!]



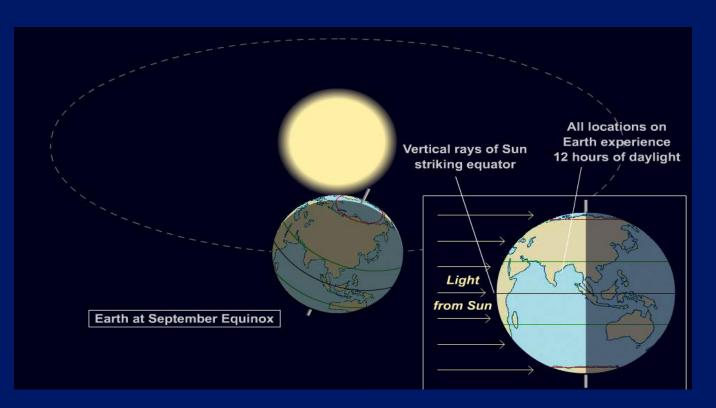
#### Preview of Topic #10: p 63

### Today is the "First Day of Fall"

#### in the Northern Hemisphere!



Google



#### You can view the animation yourself at:

http://mesoscale.agron.iastate.edu/agron206/animations/01\_EarthSun.html

World leaders urged to change course at UN climate summit

Climate change takes front and center at United Nations summit



No "Plan B" for climate action as there is no "Planet B, says UN chief

ACCHER AN PLANE & DOWN 2005



REPARATIONS for

BAN NEW

FOSSIL FUEL PROJECTS

STOP THE

CLIMATE DEBT.

STOP THE

CLIMATE CR5

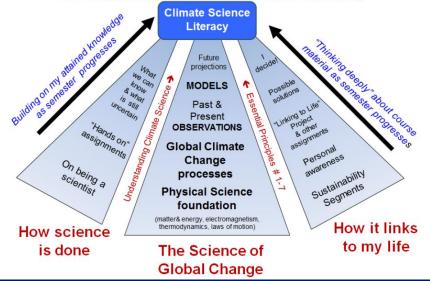




#### **Our CLASS:**

- 1. Learn & Understand the SCIENCE underlying all this!
- 2. Decide for yourself

#### GOAL: Enhanced Understanding Of Global Change Science, How It Operates, & What It Means To Me Personally

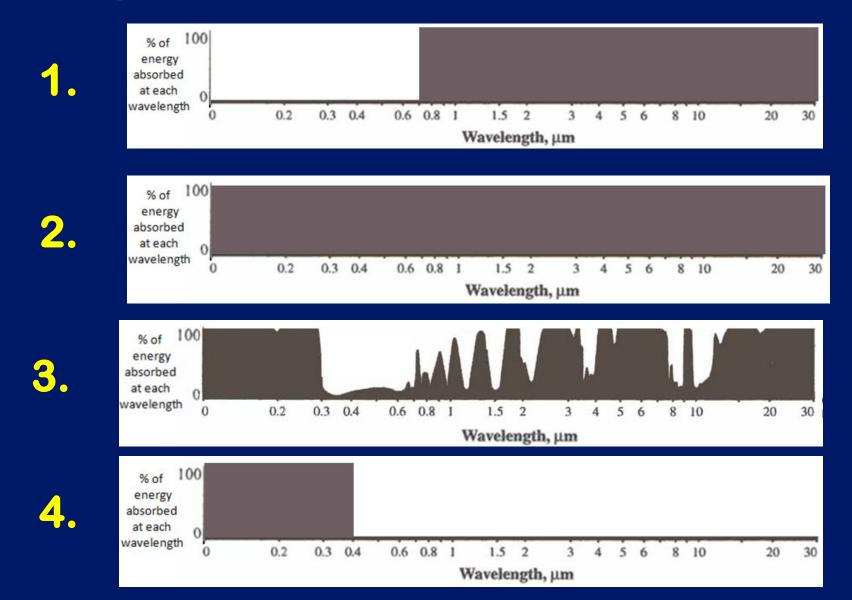


3. Link this to YOUR LIFE in the way you want to live it ....

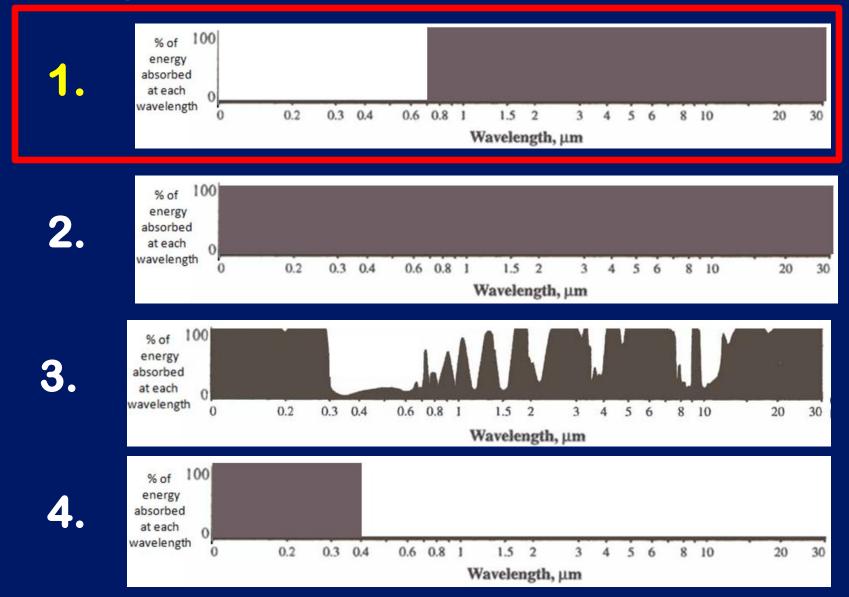
#### Sooooo... On with the SCIENCE!

Fire up your CLICKERS for some questions to solidify the concepts from the last few classes .....

# **CLICKER Q1** Which of the following absorption curves represents a <u>hypothetical</u> atmosphere that has a **"perfect" greenhouse effect ?**

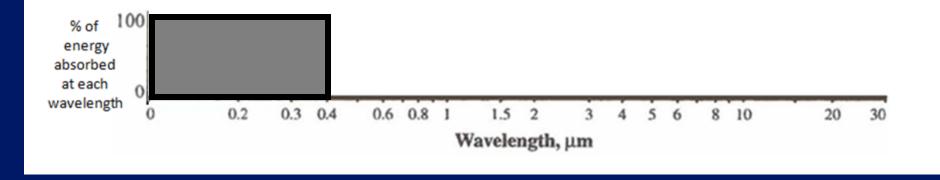


# **CLICKER Q1** Which of the following absorption curves represents a <u>hypothetical</u> atmosphere that has a **"perfect" greenhouse effect ?**

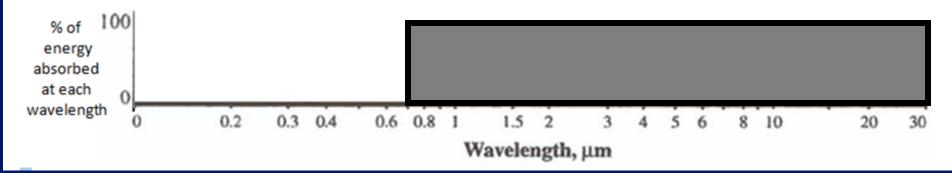


# **Next:** The ANSWERS to the first part of G-1:

Q1. Draw an absorption curve for a hypothetical gas that can absorb <u>ALL</u>UV radiation but <u>zero</u> visible light and IR radiation. Then **shade in the area under your curve** in this and subsequent questions.

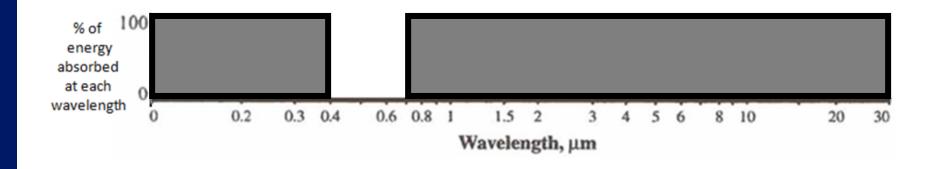


Q2. Draw an absorption curve for a "perfect" greenhouse gas that absorbs ALL IR radiation, but no visible or UV:

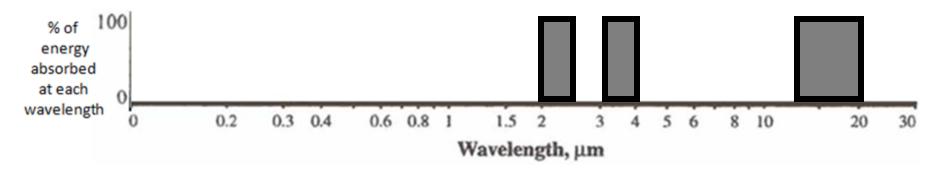


### The ANSWERS to the first part of G-1:

Q3. Draw an absorption curve for a hypothetical gas that absorbs ALL UV radiation and ALL IR radiation, but leaves a "WINDOW" open for visible light, allowing the visible light wavelengths to pass through the gas unimpeded without being absorbed:

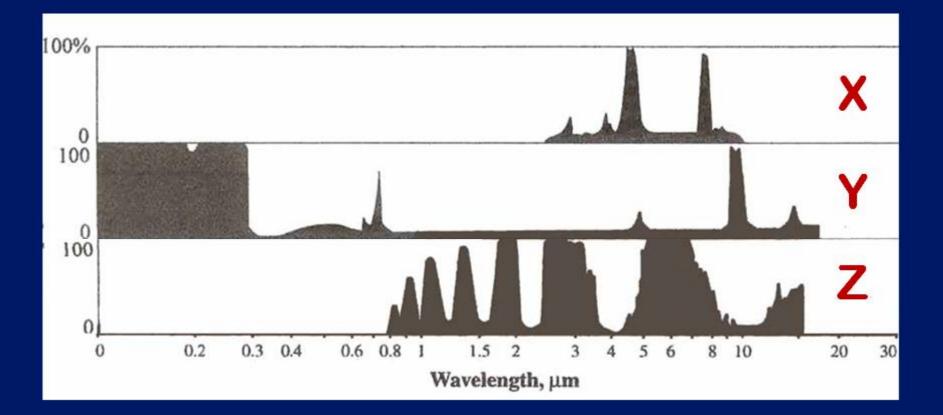


Q4. Draw an absorption curve for a hypothetical gas that can absorb 100% of the IR radiation in these three wavelength bands: band from 2 to 2.5 µm band from 3 to 4 µm band from 13 to 20 µm



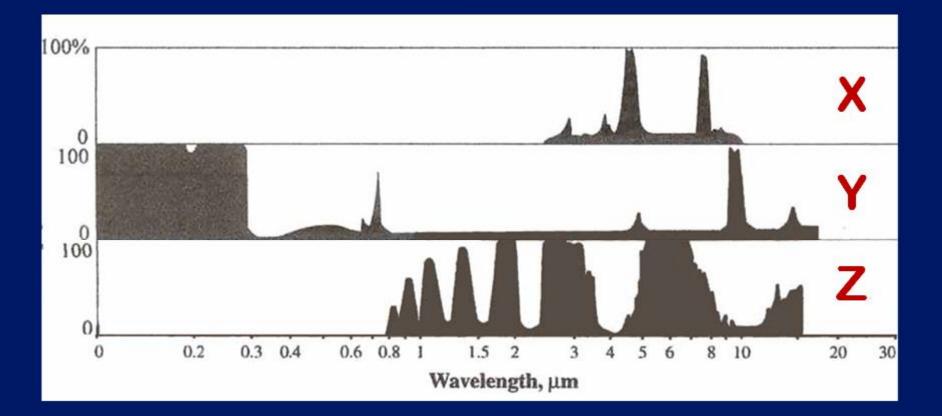
#### CLICKER Q2 – Which of the following absorption curves is for a GAS that is NOT a greenhouse gas!

#### 1: X 2: Y 3: Z 4: NONE of THEM

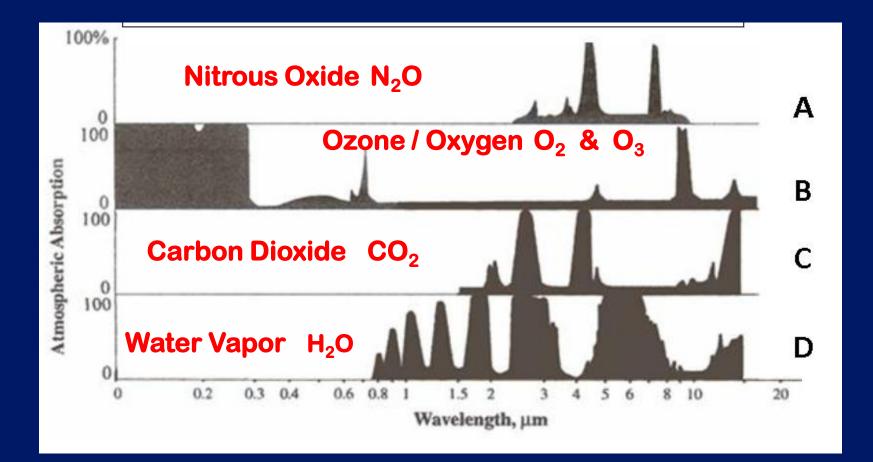


CLICKER Q2 – Which of the following absorption curves is for a GAS that is NOT a greenhouse gas!

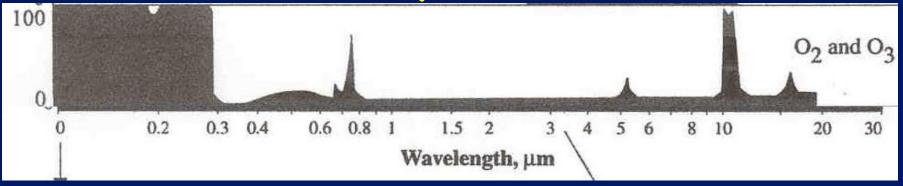
## 1: X 2: Y 3: Z 4: NONE of THEM



### The ANSWERS to the first part of G-1:



#### CLICKER Q3 HOW IS OZONE (actually O<sub>3</sub> & O<sub>2</sub>) unique???



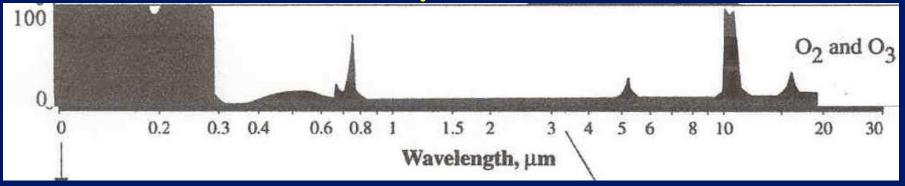
1) It absorbs only UV – hence it's <u>NOT</u> a GHG

2) It absorbs almost ALL visible wavelengths

3) It absorbs **BOTH** UV and IR so IS a GHG

4) It absorbs BOTH UV and IR so is <u>NOT</u> GHG

#### CLICKER Q3 HOW IS OZONE (actually O<sub>3</sub> & O<sub>2</sub>) unique???



1) It absorbs only UV – hence it's <u>NOT</u> a GHG

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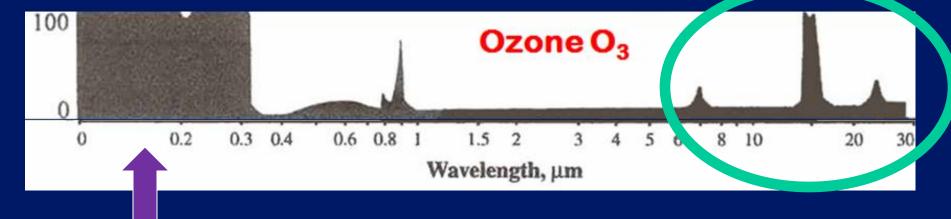
3) It absorbs **BOTH** UV and IR so IS a GHG

4) It absorbs BOTH UV and IR so is <u>NOT</u> GHG

 $\odot$ 

But only the IR absorption makes it a GHG!!

# Absorption in this part of the absorption curve (IR wavelengths) indicates that OZONE is a greenhouse gas . . .

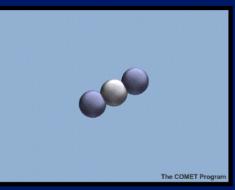


... even though OZONE also absorbs radiation in the UV part of the spectrum!

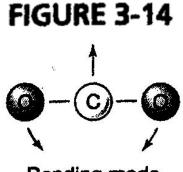
Review	Gas Water vapor (H <sub>2</sub> O)	wavelength	Primary absorption wavelengths (in micrometers)	
CH CH		0.8 1 1.5 2 to 3.5	4 to 7 9 to 10 11 to 20	
	Molecular oxygen (O <sub>2</sub> ) and Ozone (O <sub>3</sub> )	0.0001 to 0.280 8.5 to 10		
NITROUS OXIDE	Nitrous oxide (N <sub>2</sub> O)	4 to 5 7 to 7.5		
	Carbon dioxide (CO <sub>2</sub> )	2 to 2.5 3 to 4 13 to 20	4	

#### In SGC E-Text Chapt 3:

#### **IR radiation!**

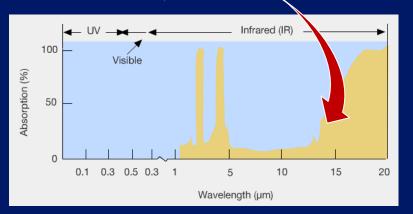


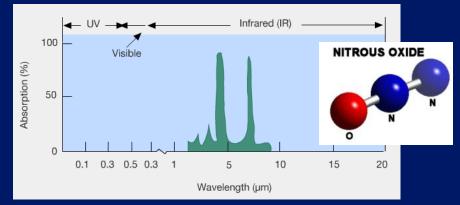
As a triatomic molecule, one way that  $CO_2$ vibrates is in a **"bending mode"** 



Bending mode (15-µm band)

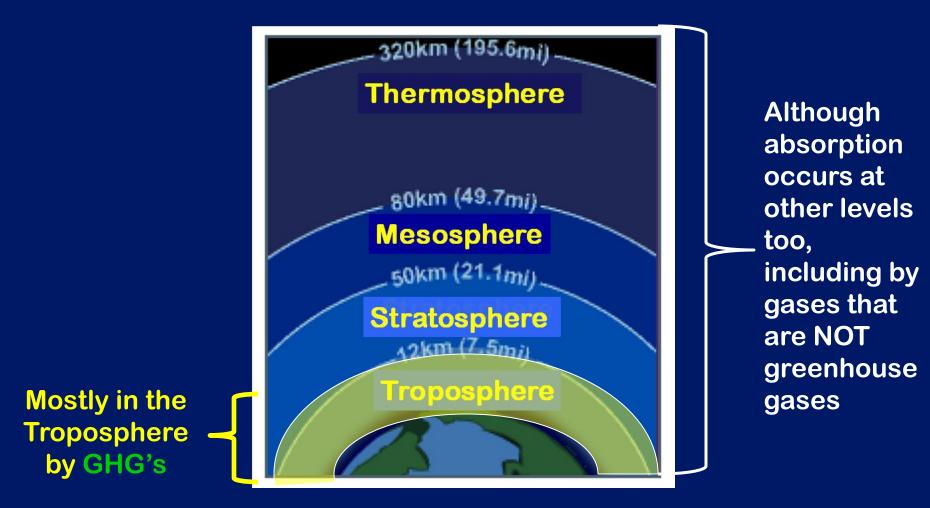
This vibration mode has a frequency that allows CO<sub>2</sub> to absorb IR radiation at a wavelength of about 15 micrometers





What about another triatomic molecule:  $N_2O$  (Nitrous oxide)?

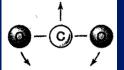
#### WHERE IS ALL THIS ABSORPTION HAPPENING?



#### more in Topic #6 on Thursday . . .

#### **CARBON DIOXIDE**

CO<sub>2</sub> moves in and out of the atmosphere as part of the

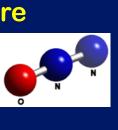


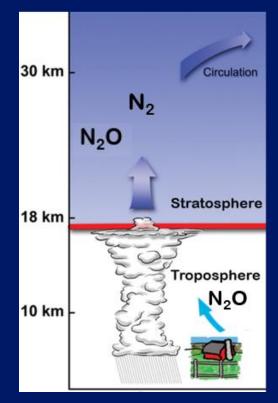
**Carbon Cycle:** 



#### **NITROUS** OXIDE

N<sub>2</sub>O moves in and out of the atmosphere as part of the Nitrogen Cycle





N<sub>2</sub>O is produced naturally in soil



also comes from fossil fuel combustion, burning forests, use of nitrogen fertilzers



# NITROUS OXIDE Another View

# DANCE YOUR PhD !!

#### DANCE YOUR PhD:



This graduate student is demonstrating the quantum behavior of a molecule of  $N_2O$ :

- one hand = a nitrogen atom
- torso = central nitrogen
- other hand = an oxygen atom

Nitrous Oxide (N<sub>2</sub>O) acts as a greenhouse gas through the absorption of radiation in 3 vibrational modes.

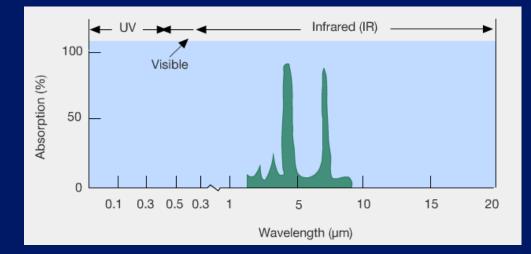
Now, 3 dancers will exhibit the 3 specific movements of N<sub>2</sub>O's vibrational modes . .

The N<sub>2</sub>O starts in the soil where it is produced by microbial activity and "moves on up" into the atmosphere.



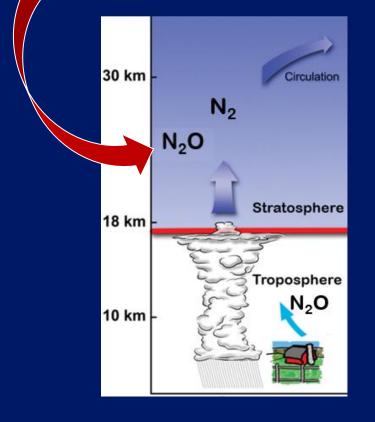
 $\odot$ 

They grads will "dance" the 3 bending modes of  $N_2O$  that are due to Infrared IR absorption at 3 different wavelengths





Stepping onto chairs represents the circulation of the N<sub>2</sub>O to higher levels in the atmosphere (the stratosphere)

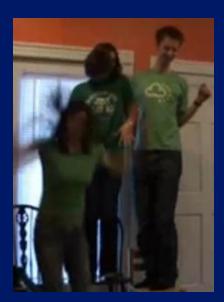


... where it is then subject to intense Ultraviolet (UV) radiation from the sun.



With the high energy from the UV radiation bombarding the  $N_2O$  the dancers go crazy with high energy dancing.

Eventually the high intensity UV radiation leads to the destruction of one of the N<sub>2</sub>O molecules (called "photolysis") Shown by jumping from the chair at the end <del>></del>



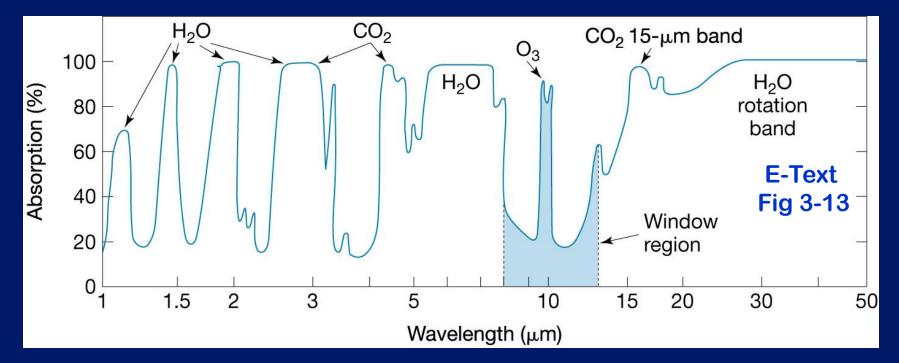


## DANCE YOUR PhD !!

http://www.youtube.com/watch?v=L5j6BS3XoLc

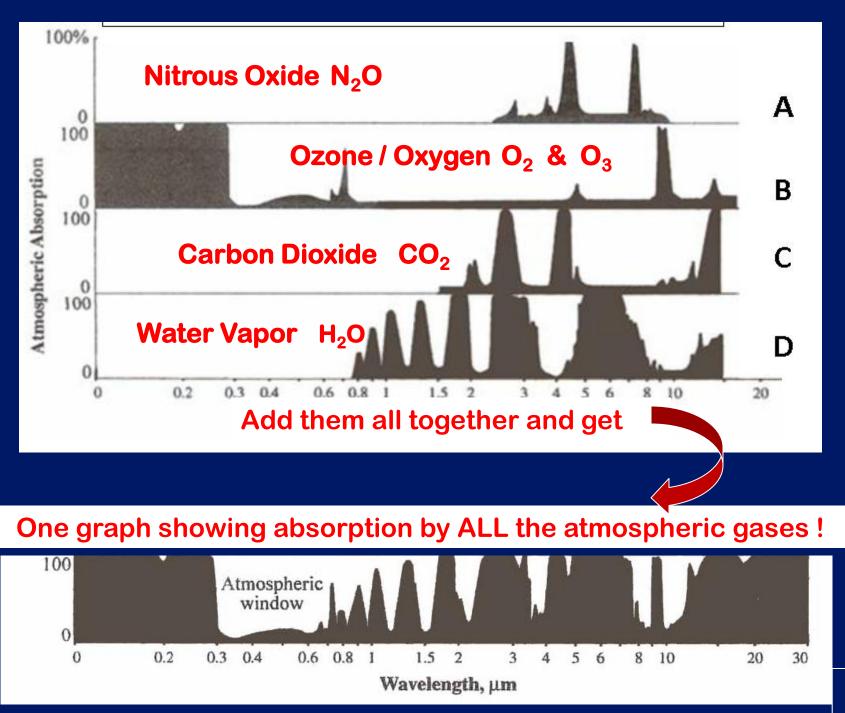
# Close up view of combined absorption of IR wavelengths by GHG's: $H_2O$ , $CO_2$ , $O_3$ :

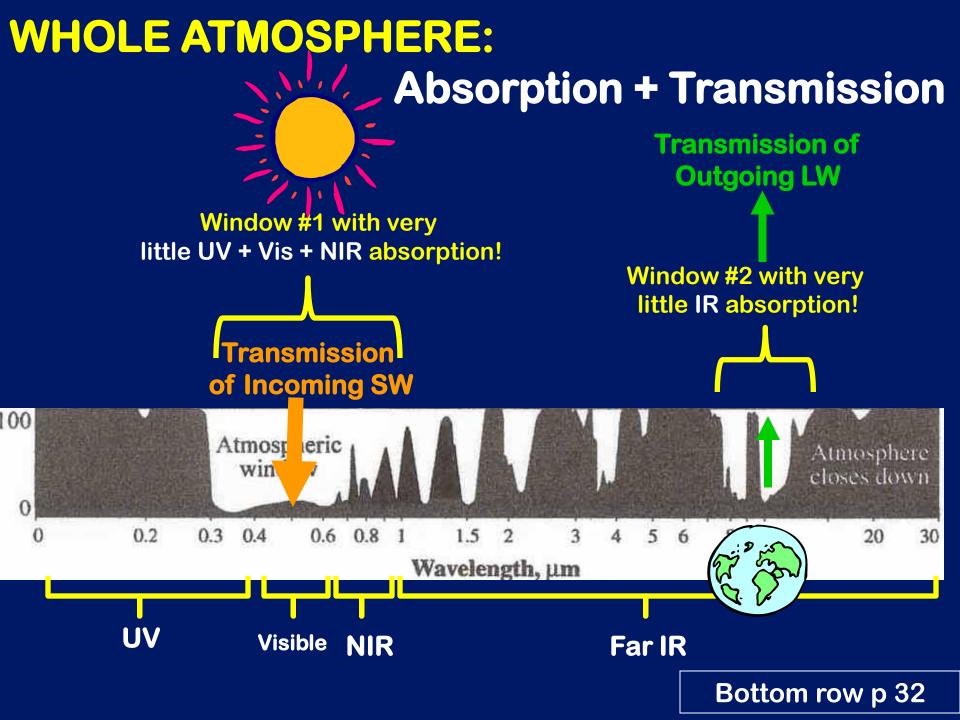




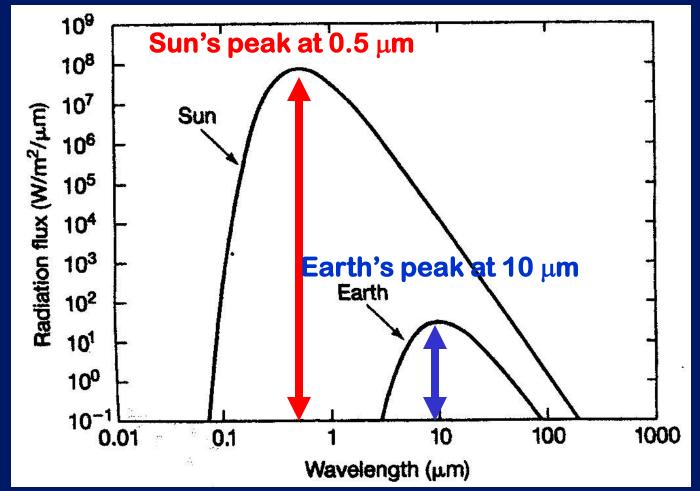
Explore Absorption Curves yourself:

#### http://apollo.lsc.vsc.edu/classes/met130/notes/c hapter2/42\_Selective\_Absorption/42.html





# IncomingOutgoingSW SOLAR (UV + Vis)LW TERRESTRIAL (IR)windowwindow



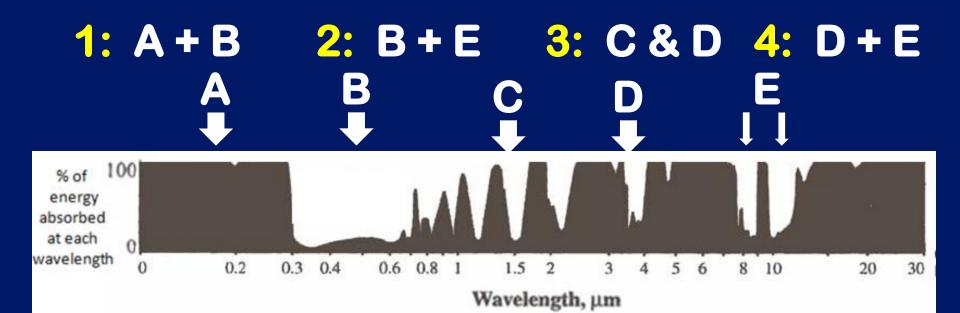
#### **REMEMBER THIS???**

Review p 28

**CLICKER Q4 -** Here's the absorption curve for ALL the gases in the atmosphere put together, i.e. curve for the "Whole Atmosphere"

We just talked about two "windows" in the curve that indicate at what wavelengths radiation easily comes IN to the surface of the Earth or escapes OUT to Space.

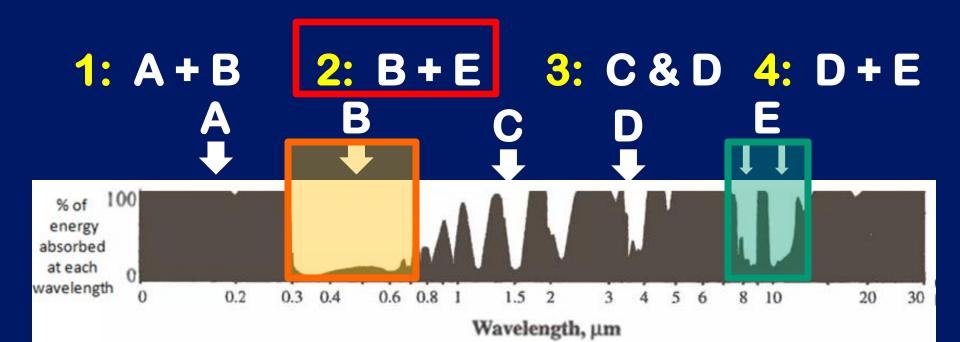
**Q.** Where are these two windows?



**CLICKER Q4 -** Here's the absorption curve for ALL the gases in the atmosphere put together, i.e. curve for the "Whole Atmosphere"

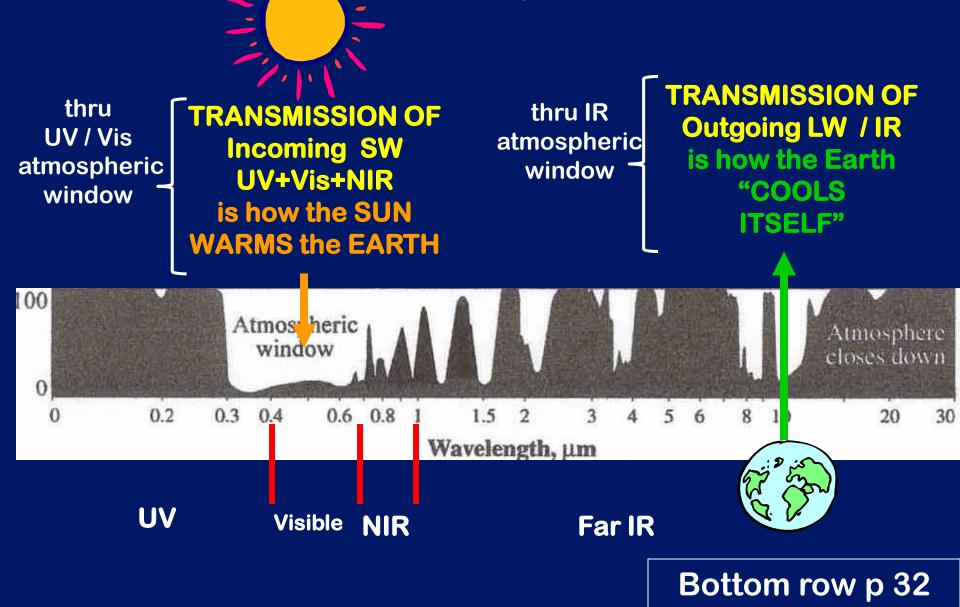
We just talked about two "windows" in the curve that indicate at what wavelengths radiation easily comes IN to the surface of the Earth or escapes OUT to Space.

**Q.** Where are these two windows?



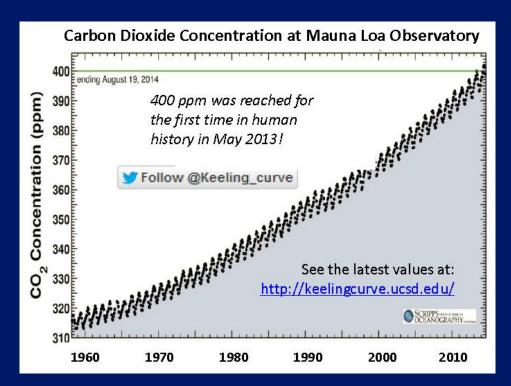
#### WHOLE ATMOSPHERE:

#### **Absorption + Transmission**



#### If the Earth can "COOL itself" by transmitting IR through the IR Window, WHY SHOULD WE BE SO CONCERNED ABOUT INCREASING CO<sub>2</sub> ?

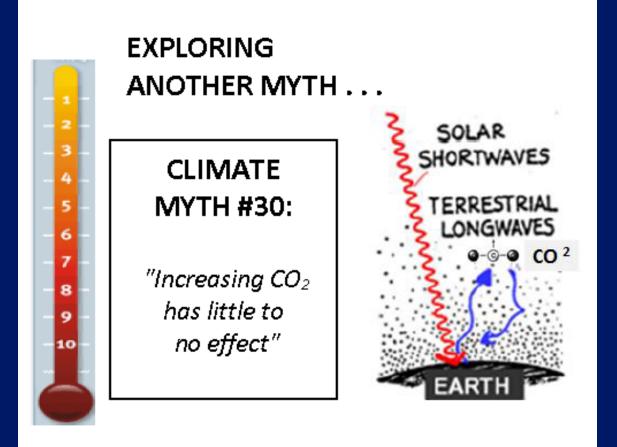
Even with the increases seen on the Keeling Curve,  $CO_2$ 's concentration in the atmosphere is really low compared to N<sub>2</sub> and O<sub>2</sub> isn't?





Thinking more deeply symbol

# WHY BE CONCERNED ABOUT INCREASING $CO_2$ ?



Bottom of p 31



#### www.skepticalscience.com

### How do we know more CO<sub>2</sub> <u>is</u> causing warming?



The skeptic argument...

"Increasing CO2 has little to no effect on enhancing the GREENHOUSE EFFECT because the amount is so small compared to the amount of other gases in the atmosphere.

Therefore the increase in human-produced CO2 (as seen in the Keeling Curve) is NOT the cause of recent global warming!!

http://www.skepticalscience.com/empirical-evidence-for-co2-enhanced-greenhouse-effect.htm

## How would you respond?



"Thinking more deeply" symbol ->

A KEY POINT to respond with is embedded in the box on "IMPLICATIONS OF LAW #6 FOR GLOBAL CLIMATE CHANGE" on p 31

Read the box ... then think a bit ... WHICH ITEMS (a - g) have relevant info for responding to this skeptic's argument?

## NOW DISCUSS IN YOUR GROUPS!!! & pick someone to respond back for your group

p3

## IT'S TIME TO WRAP IT UP AND QUIET DOWN

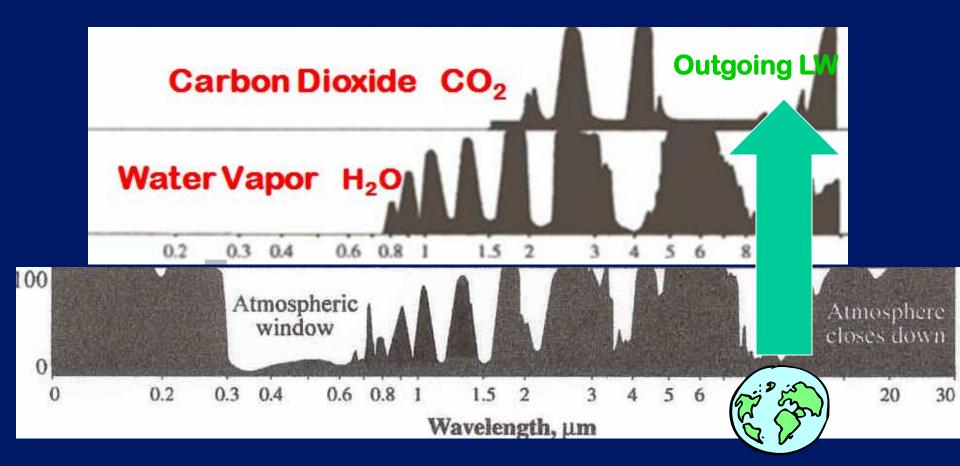


## **REPORT BACK!!!**

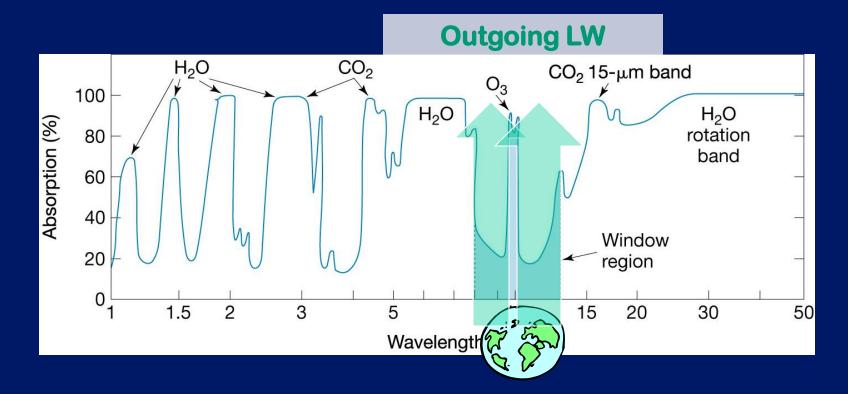
c) GREENHOUSE GASES both absorb and emit electromagnetic radiation in the infrared (IR) part of the spectrum – once IR is absorbed by the greenhouse gases in the atmosphere, it can be emitted back to the Earth's surface to heat it all over again!!

This is called the <u>GREENHOUSE EFFECT!</u>

f) Since 15 μm is close to the peak of Earth's outgoing radiation, (10 μm), this absorption band keeps a lot of Earth's longwave radiation from escaping to space. → A gas has a <u>BIG effect</u> if it absorbs in or near a "window" of wavelengths where the atmosphere is fairly transparent.



→ A gas has a <u>BIG effect</u> if it absorbs in a "window" of wavelengths where the atmosphere is fairly transparent (and the IR would otherwise escape to space!)



 $H_2O, O_3, and CO_2$ are all very close to the outgoing IR window Therefore they are effective in absorbing outgoing IR wavelengths of energy!



g) If a gas absorbs radiation of any wavelength, the amount absorbed will be proportional to

(a) the number of molecules of gas and

(b) the intensity of radiation of that wavelength.



But... is there enough volume of these "trace gases" to <u>REALLY make a difference in the Greenhouse Effect</u> and therefore increase the temperature?

**GIVE ME MORE EVIDENCE!** 

**IS** this GH Effect measurable??

#### Less IR going to space

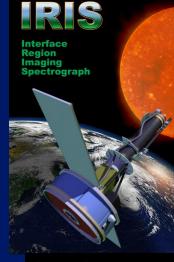
TODAY

**1970s** 

**More IR** 

radiating

downward



Interface Region Imaging Spectrograph

How to RESPOND to a SKEPTIC:

• An enhanced greenhouse effect from CO<sub>2</sub> has been confirmed by multiple lines of empirical evidence:

• Satellite measurements of infrared spectra over the past 40 years observe less energy escaping to space at the wavelengths associated with  $CO_2$ .

FOURIER TRANSFORM INFRARED SPECTROSCOPY (FTIR) ANALYSIS



• Surface measurements find more downward infrared radiation warming the planet's surface.

This provides a direct, observed & measured CAUSAL LINK between CO<sub>2</sub> and global warming.

OK .... TIME NOW FOR GROUPS TO FINISH THE LAST PART OF G-1 SEE YOU THURSDAY!!!